PHOTO ELECTRON EMISSION MICROSCOPY (PEEM), FIELD EMISSION ELECTRON MICROSCOPY (FEEM) AND THERMIONIC FEEM OF NITROGEN AND SULFUR DOPED CARBON FILMS.

F.A.M. Köck¹, S. Gupta², G. Morell³, B. R. Weiner², J.M. Garguilo¹, Billyde Brown¹, and R.J. Nemanich¹
[1]Department of Physics
North Carolina State University
Raleigh, NC 27695-8202
[2] University of Puerto Rico
Dept Phys
POB 23343, San Juan, PR 00931
[3] University of Puerto Rico

Dept Phys Sci

POB 23323, Rio Piedras, PR 00931

We have studied Nitrogen doped diamond films and Sulfur doped Carbon Films as low temperature thermionic-field electron sources. Nitrogen doped Diamond Films have been synthesized by Plasma Enhanced CVD on Si (100) and molybdenum substrates. Sulfur doped films have been prepared by HFCVD on molybdenum substrates. [1, 2] Raman Spectroscopy for the Nitrogen doped Films shows a peak at 1340cm⁻¹ characteristic for diamond and a broadened peak at 1554cm⁻¹ corresponding to sp² bonded carbon. Sulfur doped films show two broad peaks at 1354cm⁻¹ and 1590cm⁻¹ indicating nanocrystalline diamond. Photo Electron Emission Microscopy shows a uniform electron emission at room temperature for the nitrogen doped diamond films in contrast to a strongly localized electron emission for the sulfur doped films. Surface treatment of the nitrogen doped diamond films enhances electron emission at elevated temperatures starting at around 600°C that increases with increasing temperature. Hydrogen and Titanium passivation induce a negative electron affinity to lower the energy barrier for electrons to escape into the vacuum. [3] The degradation of the Hydrogen layer starts at $\sim 720^{\circ}$ C in contrast to the Titanium layer that is stable up to 950°C. At elevated temperatures the electron emission for the nitrogen doped films as imaged with the Electron Emission Microscope shows a uniform distribution over a wide surface region. Sulfur doped carbon films with electron emission originating mainly from 'hot spots' on the surface at room temperature show an increase in electron emission from those localized regions at temperatures as low as 340°C. Nitrogen introduces a single substitutional donor level at 1.6eV below the CBM, as well as the incorporation of other defect levels. [4, 5] These defect states may play an important role in enhancing electron emission especially at high fields. The energy levels induced by sulfur in diamond are calculated to be 0.15 and 0.5eV from the bottom of the conduction band, for S-0 and the singly ionized state S+, respectively. [6] The contribution of different states to the electron emission for the nitrogen and sulfur doped films has to be investigated.

We would like to acknowledge Ms S. Gupta and Dr. G. Morell from the University of Puerto Rico for preparation of the sulfur doped films.

[1]. S. Gupta, B. L. Weiss, B. R. Weiner and G. Morell, Appl. Phys. Lett. 79, 3446 (2001)

[2]. S. Gupta, B. R. Weiner and G. Morell, Appl. Phys. Lett. 80, 18 Feb. issue (2002)

[3]. R.J. Nemanich, P.K. Baumann, M.C. Benjamin, O.-

H. Nam, A.T. Sowers, B.L. Ward, H. Ade and R.F. Davis, Appl. Surface Sci. 130-132 (1998) 694-703

[4]. E. Rohrer, C.F.O. Graeff, R. Janssen, C.E. Nebel, and M. Stutzmann, Phys. Rev. B **54** (1996) 7874-7880

- [5]. J.B. Cui, J. Ristein, M. Stammler, K. Janischowsky,
- G. Kleber, L. Ley, Diamond and Related Materials 2000, 9:3-6:1143-1147

[6]. Saada D, Adler J, Kalish R, Applied Physics Letters 77 (6): 878-879 AUG 7 2000



Figure 1 Thermionic FEEM of Nitrogen doped, H-passivated diamond at 725°C.



Figure 2 Thermionic FEEM of Sulfur doped nanocrystalline diamond at 340°C.