## Photophysical and Photochemical Properties of C<sub>60</sub>-linked Dendritic Multi-zincporphyrin Arrays

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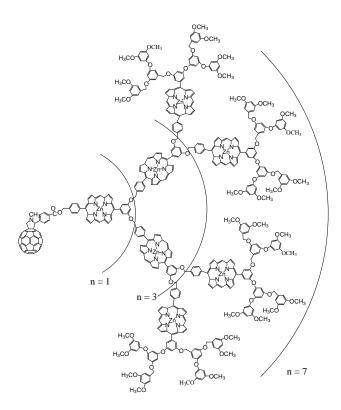
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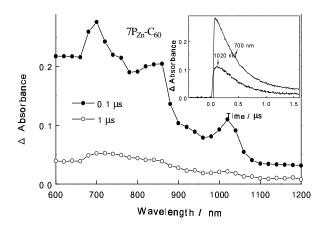
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Photochemical and photophysical properties of C<sub>60</sub>-linked dendritic multi-zincporpyrins (scheme 1) investigated by measuring been fluorescence lifetimes and transient absorption spectra with changing the generation of dendritic multi-zincporphyrins. Fluorescence lifetimes of the zincporphyrin moiety become short in various solvents, suggesting that intramolecular electron-transfer takes place from the singlet excited states of zincporphyrin moiety to the connected C<sub>60</sub> moiety. The fluorescence lifetimes of 1Pzn-C60 are shorter than those of  $3P_{Zn}$ - $C_{60}$  and  $7P_{Zn}$ - $C_{60}$ , in which latter two nP<sub>Zn</sub>-C<sub>60</sub> showed similar fluorescence lifetimes, suggesting that the charge-separation takes place from the singlet excited states of zincporphyrin moiety at the first- and second-generations. In the charege-separation rate constants and the quantum yields are almost similar with changing solvent polarity from anisole and THF to benzonitrile.

In anisole, absorption of the radical anion of the C<sub>60</sub> moiety decayed quickly during the ns-laser pulse, showing the lifetimes of the radical ion-pair are as short as ca. 10 ns. In THF and benzonitrile, slow rises of the radical anion of the C<sub>60</sub> moiety were observed in the nanosecond time scale. In such case, the charge-recombination rates of the radical ion-pair states are as slow as ca.10<sup>6</sup> s<sup>-1</sup>. The longest lifetime of the radical ion-pair was observed for 7P<sub>Zn</sub>-C<sub>60</sub> in benzonitrile (ca. 500 ns) compared with  $3P_{Zn}$ - $C_{60}$  (ca. 400 ns) and  $3P_{Zn}$ - $C_{60}$  (ca. 400 ns). The lifetimes of the ion-pairs in THF are slightly shorter than those in benzonitrile. The activation energies of the slow charge-recombination process were evaluated to be ca. 7, ca. 8, and ca. 10 kcal mol<sup>-1</sup> for first-, second-, and third-generations, respectively. The higher activation energy may be caused by the delocalization of the radical cation (hole) in whole prophyrin moieties even in 7P<sub>Zn</sub>-C<sub>60</sub>.



**Scheme 1.**  $C_{60}$ -linked dendritic multiporphyrin arrays ( $nP_{Zn}$  - $C_{60}$ ).



**Figure 1.** Nano-second transient absorption spectra of  $7P_{Zn}$ - $C_{60}$  in benzonitrile observed by the laser excitation with 532 nm light in Ar-saturated solution. Inset: Time profiles at 700 nm for triplet state of and at 1000 nm of radical anion of  $C_{60}$  moiety.

## References

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