

**ELECTROCHEMICAL CONVERSION of BROWN  
COAL (Victoria Brown Coal) into SOLUBABLE  
FUMIC ACID DERIVATIVES BY USING SMALL  
ZAPPI CELL FOR  
BIO-GASIFICATION PROCESS**

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For the production of methane gas from low rank coal, the combination of two different technologies such as electrooxidation and fermentation processes has been investigated. First, low rank coals are oxidatively converted into soluble fuming acid derivatives which was further allowed to convert into methane gas by employing a

methane gas fermentation process. The objective of the work conducted at ICC was the electrooxidative conversion of samples of coal into an electrooxidatively more-functionalized coal suspension, which undergoes further fermentation operation in different laboratory. The electrochemical oxidation of the low rank coal was carried out by using the Zappi cell supplied by the Electro-synthesis Company, Inc. in USA. The above projects have been attempted by the partnership between Nippon Kokan Techno Service Co., Ltd., The Tokyo Institute of Technology, The Institute of Creative Chemistry, and The Electro-synthesis Company, Inc. sponsored by NEDO (New Energy and Development Organization) of Japan.

**Outline of Research Plans**

1. Confirmation of Electrolysis Conditions in Zappi-Cell
2. Choice of Electrolytes for Low Rank Coal (LRC) in Zappi-Cell
3. Effect of pH under Electrooxidation of LRC with Zappi-Cell
4. Oxidation Efficiency in Zappi-Cell under Different Electrodes
5. Factors for Generation of Oxidatively Active Species
5. Novel Electrooxidation System for Low Rank Coal by Zappi-Cell

The modification of the coal proceeds by an electrooxidative process: generally, the oxidation may occur on the surface of the anode electrode (direct oxidation) or by the action of a mediator that is produced at the anode and then oxidizes the coal (indirect oxidation). Typical mediators could be halide salts, for example. Other oxidizing species are those produced by oxidation of water, and they include: oxygen, O<sub>2</sub>; ozone, O<sub>3</sub>; hydrogen peroxide, H<sub>2</sub>O<sub>2</sub>; hydroxyl radicals, OH

**Experimental Section**

A commercially available Zappi Cell with ca. 70 cm<sup>2</sup> total anode area was used. The Zappi Cell is an open-design electrolysis cell, patented by The Electro-synthesis Company. Typical electrolysis conditions are presented in the following Table. The detail results are discussed in the meeting.

Electrooxidation of Victoria Low Rank Coal

Experiment No.	Run 1	Run 2	Run 3
quantity g	5.04	5.08	9.92
Pretreatment	Non-Treatment	Mortar Atomization	Mortar Atomization
Water Used L	1	1	1
Electrolyte			
KH <sub>2</sub> PO <sub>4</sub> g	2.5	2.5	2.5
Na <sub>2</sub> HPO <sub>4</sub> g	5.0	5.0	5.0
Voltage V	6.9~10.0	6.2~8.6	7.1~9.2
Current A	24.2~30.0	30.0	30.0
Reaction Time h	9.0	6.0	7.0
pH	6.95~7.26	7.04~7.21	6.84~7.12
Temp. °C	27.7~58.1	24.2~53.6	25.2~54.1
Style of Sample	Solid Foam Suspension	Solid Foam Suspension	1L PE bottle aq. solution
Remarks			

i) Suspended Low R. Coal solution : 5g/L

ii) Atomization is performed in mortar.