## STABLE TiO<sub>X</sub> SUB-MICROMETER CHANELLS

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Channel architectures of sub-micrometer sizes have potential applications in many fields [1]. In a previous work [2], stable, hollow sub-micrometric structures were patterned in films deposited by PECVD. The obtention of sub-micrometer, stable channels of r.f. sputtered titanium oxide films with different stoichiometries is described in this work. The structure was obtained by holographic exposing a photoresist, depositing the film and subsequentely dissolving the photoresist.

The AZ1518 photoresist was spin coated onto glass substrates, of dimensions 2.5 vs 2.5 cm. The samples were pre-beaked for 20 min at 70 °C and then exposed in a stabilized holographic setup using the 458 nm line of an Ar laser. Masks with patterns of  $0.3\mu m$ height and a period of 1µm were used. After exposure the samples were developed in AZ 351 diluted in de-ionized water. In sequence, films of titanium oxide were deposited by r.f. magnetron sputtering from a titanium target, under an O<sub>2</sub>-Ar atmosphere. \in order to obtain films with different stoichiometries, both pressure and power were maintained at constant values, and the O2 flow ( $\phi$ ) was varied by means of a controlled flowmeter. Thickness of the films were around 100 nm, as measured by profilometry. Finally, for removal of the photoresist, the samples were immersed in acetone.

Analysis by X-ray diffraction showed that all films were amorphous (or nanocrystalline). The spectral transmittance data are consistent with films presenting distinct stoichiometries.

Figure 1 show the SEM photographs for films deposited under different  $O_2$  flow. The submicrometer channels were stable for all samples, in the whole extension of the mask, and present uniform walls, particularly at low  $\phi$ . Preliminary work performed with thinner films (~10 nm) indicate that this stability is not maintaned for all thickness.

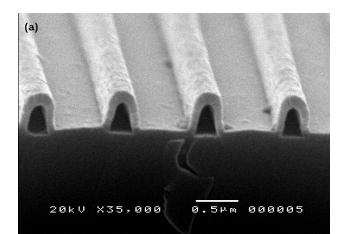
Samples were also deposited onto conducting substrates (ITO/glass), with excellent results (Fig. 2). Since TiOx is a well-known electrochromic material [3], this result open the possibility of exploring optoelectrochemical devices with optimized performance.

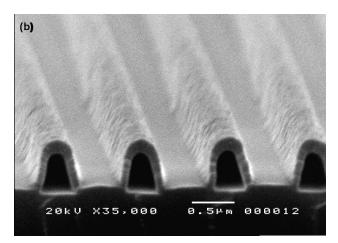
## References

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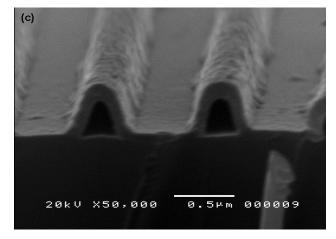


Figure 1 - SEM photographs of  $TiO_x$  samples deposited at different O<sub>2</sub> flows ( $\phi$ ). (a)  $\phi$ =1.0 sccm; (b)  $\phi$ =1.6 sccm; (c)  $\phi$ = 2.2 sccm

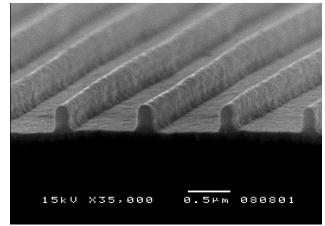


Figure 2 - SEM photograph of  $TiO_x$  sample deposited onto ITO/glass substrate.