Epitaxial Electrodeposition of Cuprous Oxide onto Single Crystal Silicon(001)

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Epitaxial films of cuprous oxide, Cu2O, are electrodeposited from aqueous solution onto singlecrystalline Si(001). The epitaxial growth of Cu₂O onto Si is an unexpected result, because there is a strong driving force for the oxidation of Si to form a native amorphous SiO₂ layer. The Cu₂O films are single-crystalline in nature, with little or no fiber texture. High resolution Xray diffraction shows that the very large lattice mismatch of -21.4% is reduced to +11.2% by the formation of a Cu₂O(001)[100]//Si(001)[110] orientation relationship, in which the film is rotated 45° around the common [001] axis. Cross-sectional transmission electron microscopy studies suggest that the growth mechanism involves an initial electroless deposition of Cu2O epitaxial seeds in direct contact with the Si(001), with the concomitant oxidation of Si to form a layer of amorphous SiO₂ about 3 nm thick. Electrochemical deposition of Cu₂O then proceeds on the epitaxial seeds, with the deposit growing not only perpendicular to the surface but also laterally across the SiO₂ interlayer. This type of lateral overgrowth is used in the production of semiconductor-on-insulator (SOI) structures to avoid defects due to misfit strains.

The epitaxial films were deposited onto p-type Si(001) from an aqueous solution of 0.4 M Cu(II) and 3 M lactate ions at pH 9 using a bath temperature of 65° C. The wafers were supplied by Virginia Semiconductor doped with boron to a resistivity of 7.5 ohm-cm. The wafers were degreased in ethanol and acetone and then rinsed with HPLC water before etching to produce an H-terminated surface. The etch consisted of 5% HF for 1 min, hot HPLC water for 15 min, and 5% HF for 10 s, followed by a thorough washing with HPLC water. Ohmic contacts were made with GaIn eutectic.

A plan-view SEM image of a film that is approximately 2 µm thick is shown in Fig. 1. The inplane order is evident in the SEM image. The Bragg-Bentano scan in Fig. 2a probes the out of plane orientation of the film. Only the (002) and (004) peaks of Cu₂O are observed. Fig. 2b shows a (220) pole figure for the film. The radial grid lines in the pole figure correspond to 30° increments in the tilt angle. Four sharp spots are seen at a tilt angle of 45°, corresponding to the angle between {100} and {220} planes in a cubic system. The azimuthal scans in Fig. 2c were obtained by selecting the Cu₂O(111) planes at 2θ = 36.418° and the Si(111) planes at $2\theta = 28.433^{\circ}$ and tilting the sample at a tilt angle of 54.7°. The azimuthal scans show that the film is rotated 45° about the [001] axis relative to the Si substrate.

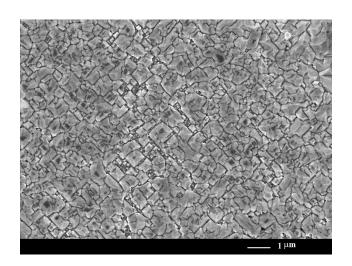
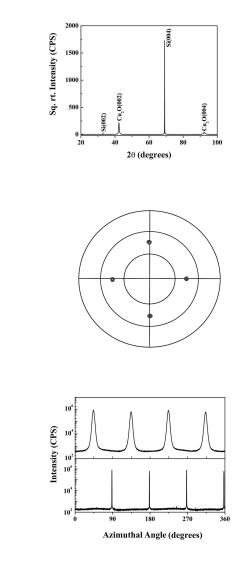


Fig. 1. SEM image of Cu_2O on Si(001)



b

Fig. 2. X-ray diffraction results for Cu_2O on Si(001)