Influence of surface defects on the metal electrochemical deposition onto *p*-type Si

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Since the recent incorporation of electrodeposited copper into electronic devices, a renewal of interest for electrodeposition and related technologies have found new applications in electronics manufacturing, especially for packaging and magnetic recording [1]. Due to the Schottky diode behavior of the semiconductor/ electrolyte junction n-type semiconductors are preferentially used in electrodeposition processes. Mechanisms and kinetics of the metal electrochemical deposition onto n-Si was recently reported (see e.g. Ref [2]).

More recently, selective metal electrodeposition onto p-Si at surface defects created by focussed ion implantation has shown that surface defects play crucial role in deposition processes [3].

The present work investigate therefore the influence of controlled mechanical defects onto the metal electrodeposition processes. The micro- and nano-scaled defects are created using a micro-indenter or an atomic force microscope (AFM) equipped with single-crystalline diamond tips. Traditional electrochemical techniques as well as micro-capillary experiments (Fig. 1) and capacitance measurements are used to compare the electrochemical behavior of intact and defective p-Si surfaces. From these investigations, it appears clearly that deposition of metals such as Cu and Pd is strongly enhanced by the introduction of mechanical defects. A patterning technique is therefore proposed (Fig. 2).

References:

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- [2] G. Oskam, J. G. Long, A. Natarajan, P. C. Searson, J. Phys. D. :Appl. Phys., 31, 1927 (1998).
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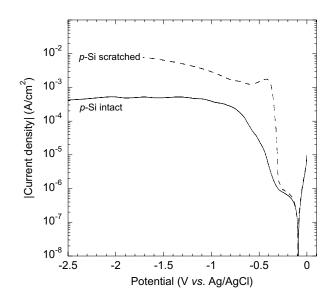


Fig. 1: Polarization curves for intact (full line) and scratched (broken line) *p*-Si obtained using a microcapillary in 0.01 M CuSO₄ + 0.05 M H_2SO_4 .

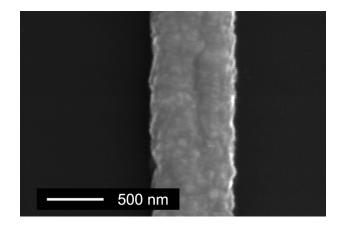


Fig. 2: SEM image of copper deposit on AFM-scratched *p*-type silicon. Deposition was carried out from $CuSO_4 (0.01 \text{ M}) + H_2SO_4 (0.05 \text{ M}) \text{ at} - 400 \text{ mV} (Ag/AgCl) during 15 s.$

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