

ANALYTICAL CHARACTERIZATION OF PROCESS PARAMETER INFLUENCE ON THE INITIAL GROWTH AND CRYSTALLINITY OF ATOMIC LAYER DEPOSITION HfO₂ THIN FILM

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Aggressive scaling down of silicon-based microelectronic devices leads to consider HfO₂ as a serious alternative dielectric gate to replace SiO₂. Crystallinity of Atomic Layer Deposition (ALD) ZrO₂ films has been extensively studied (1) but there are only few data about ALD HfO₂ thin films and effect of impurities on their final crystallinity (2). Therefore, this paper deals with thin HfO₂ films deposited on SiO₂ by ALD in a Pulsar 2000TM with HfCl₄ and H₂O, as precursors, at different temperatures (250°C to 350°C). Total X-Ray Fluorescence (TXRF) is performed to investigate the early stage of HfO₂ growth by measuring Hf and Cl content after each deposition cycle. Then, influence of process parameters and chlorine contamination brought by HfCl₄ on final crystallinity is investigated by Attenuated Total Reflection (ATR).

TXRF displays that the first deposition cycle leads to a high adsorption of Hf (a few 10¹⁴ at/cm²) whereas during the following cycles a permanent growth with a lower rate of about 4.10¹³ at/cm² is established (Figure 1). Chlorine follows the same trend and exhibits a higher chlorine concentration at the SiO₂/HfO₂ interface (Figures 2 and 3). We also evidence that Hf and Cl adsorptions are well linked and decrease when deposition temperature increases (Figures 1, 2 and 3). This means that chlorine contamination is directly correlated with the hydroxyl group coverage of SiO₂.

The influence of deposition temperature on HfO₂ final crystallinity is shown on ATR spectra of 8 nm HfO₂ oxide films grown on a specific SiO₂ surface (Figure 4): a large peak between 650 and 800 cm⁻¹ reveals the presence of amorphous HfO₂ at 300°C whereas two distinct thinner major peaks at 675 cm⁻¹ and 775 cm⁻¹ stand for crystalline HfO₂ at 350°C. The same 8 nm films grown at 350°C on other SiO₂ surface preparations lead to amorphous HfO₂. Since Cl contamination at the interface and inside HfO₂ film is higher at 300°C, Cl content both at the interface and in the film could have an influence on HfO₂ crystallinity. To refine this study, we anneal at 350°C HfO₂ films deposited at 300°C that have 10 times more chlorine in the HfO₂ volume (Figure 3). As those films remain amorphous after annealing time longer than the deposition time at 350°C, we demonstrate the major effect of chlorine in crystallization (Figure 5).

We conclude that 350°C is a transition temperature that allows to deposit crystallized 8 nm HfO₂ films by careful optimisation of process parameters and SiO₂ surface preparation. Moreover, we demonstrate the inhibition effect of chlorine content on thin HfO₂ film crystallinity.

1. C. Zhao et al., *Microelectronics Reliability*, **41**, 995 (2001)

2. J. Aarik et al., *Thin Solid Films*, **340**, 110 (1999)

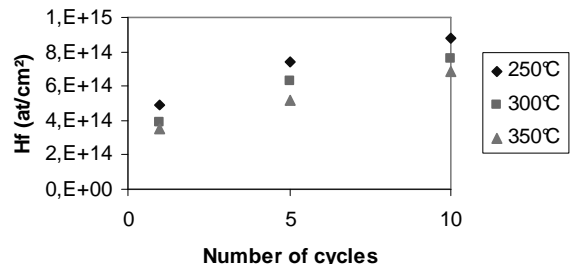


Figure 1. Influence of deposition temperature on Hf adsorption measured by TXRF on SiO₂ surface

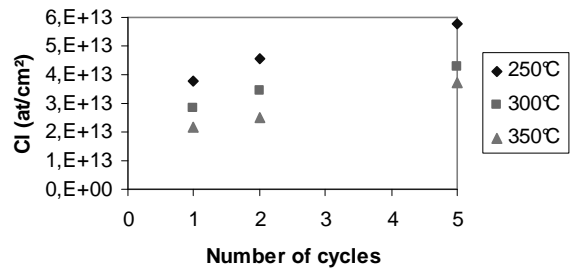


Figure 2. Influence of deposition temperature on Cl adsorption measured by TXRF on SiO₂ surface

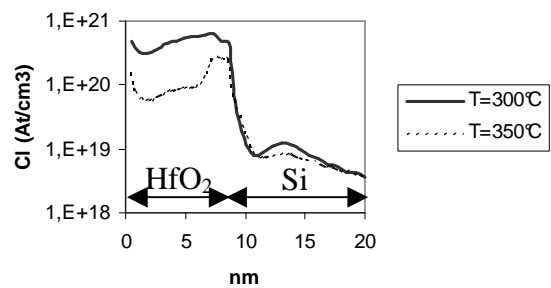


Figure 3. Influence of deposition temperature on Cl content measured by Secondary Ion Mass Spectroscopy (SIMS) for 8 nm HfO₂ layers deposited on SiO₂ surface

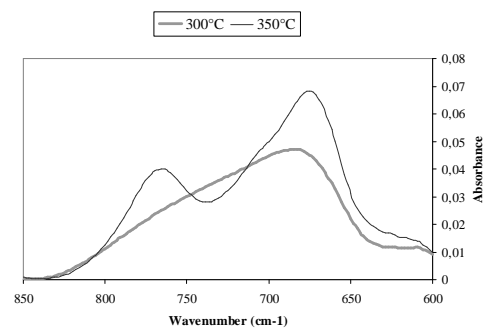


Figure 4. Influence of deposition temperature on HfO₂ crystallinity measured by ATR for 8 nm HfO₂ films deposited on specific SiO₂ surface

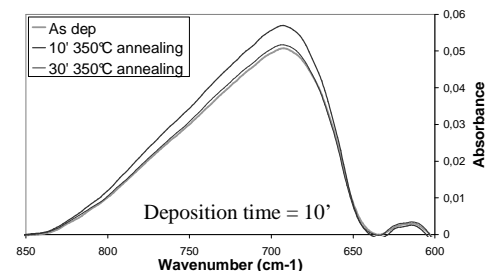


Figure 5. Effect of 350°C annealing on HfO₂ crystallinity measured by ATR for 8 nm films deposited at 300°C