

## Synthesis of Open-Cage Fullerene Derivatives and Encapsulation of Small Molecules

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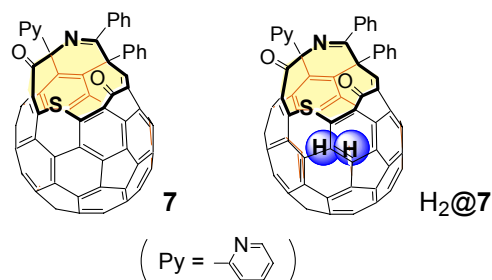
Preparation of open-cage fullerenes is of great importance in a challenging task aiming at the organic synthesis of endohedral fullerene complexes.

Previously, we found that a thermal reaction of C<sub>60</sub> with various nitrogen-containing aromatic compounds can lead to the formation of open-cage fullerene derivatives having an 8-membered-ring orifice such as **1**,<sup>1</sup> **2**,<sup>2</sup> and **3**.<sup>3</sup> The size of the orifice can be further enlarged to a 12-membered ring by aerobic oxidative cleavage of a double bond at the rim of the orifice under photochemical conditions, thus giving **4**,<sup>4</sup> **5**,<sup>2</sup> and **6**.<sup>3</sup>

Then, we found that elemental sulfur can be inserted into the rim of the orifice of compound **6** quite effectively by heating **6** in the presence of an electron acceptor such as tetrakis(dimethylamino)ethylene. Thus a novel open-cage fullerene derivative **7** with a 13-membered-ring orifice was obtained in 77% yield.<sup>3</sup> The structures of **3** and **7** were unambiguously determined by X-ray crystallography.

Theoretical calculations indicated that the energy barrier for insertion of small molecules such as neon and hydrogen should be rather low, i.e. 26.2 and 30.1 kcal/mol respectively. Actually, a remarkably large amount of hydrogen was found to be encapsulated in the derivative **7** under high pressure (800 atm) and high temperature (200 °C) conditions. The <sup>1</sup>H NMR chemical shift of the encapsulated hydrogen was observed at such a high field as -7.25 ppm due to strong shielding effects of the fullerene cage.

The detailed properties of H<sub>2</sub>@**7** will be reported together with the results on the encapsulation of smaller molecules such as rare-gases.



### References:

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