

Gibbs energy treatment on a set of $C_{60}F_{36}$ isomers

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$C_{60}F_{36}$ is known to be formed by several isomers. The computations show that the isomers are relatively close in potential energy. Moreover, the system is prepared at higher temperatures. In overall, $C_{60}F_{36}$ represents an interesting candidate for computational treatment not only in the terms of potential energy or heats of formation but also in the terms of Gibbs function. The treatment is reported, based on the PM3 semiempirical quantum-chemical method.