## $\label{eq:constraint} \mbox{Effect of Solvent on Growth of Ru and RuO_2 Films by } \\ \mbox{Liquid Injection MOCVD}$

Karol Frohlich,<sup>1</sup> Kristina Husekova,<sup>1</sup> Daniel Machajdik,<sup>1</sup> Jan Soltys,<sup>1</sup> Vladimir Patoprsty,<sup>2</sup> Peter Baumann,<sup>3</sup> Johannes Lindner<sup>3</sup> and Marcus Schumacher<sup>3</sup>

> <sup>1</sup>Institute of Electrical Engineering, SAS Dúbravská cesta 9 Bratislava 842 39 Slovak Republic

> > <sup>2</sup>Institute of Chemistry, SAS Dúbravská cesta 9 Bratislava 842 38 Slovak Republic

> > > <sup>3</sup>AIXTRON AG Kackertstr. 15-17 Aachen D-52072 Germany

We have grown Ru and  $RuO_2$  thin films by liquid injection metal organic chemical vapor deposition (MOCVD) technique. Ru (thd)<sub>2</sub>(cod) precursor dissolved in an organic solvent was injected into the low pressure MOCVD reactor. Various solvents such as isooctane, n-octane, cyclohexane, xylene, n-propanol, dioxane and 2-methoxyethyl ether, (diglyme) were used. Depending on particular solvent and deposition conditions we were able to grow either RuO<sub>2</sub> or Ru films.

Surprisingly, deposition using chemically similar solvents, isooctane and n-octane, resulted in growth of  $RuO_2$  and Ru films, respectively. Using infrared spectroscopy of reaction products we have determined different decomposition pathways for these solvents. In particular, decomposition of n-octane involves combustion in large extent and, consequently, decreases substantially partial pressure of oxygen in the reaction atmosphere. We point out that oxygen partial pressure is of crucial importance for  $RuO_2$  phase growth.