## Synthetic Studies on Bioactive Natural Products Employing Electrochemical Methodology, Synthesis of Heliannuol E

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A considerable amount of synthetic methodologies accumlated during the last decade, have been able to construct a variety of complicated organic molecules. Among such investigations, phenol-derived bioactive substances are of interest, from viewpoints of complicated aryl and alkyl-mixed frameworks, which provide diverse biological activities. In this context, we investigated the anodic oxidation chemistry of halogenated phenols to synthesize isodityrosine-class natural products (ref. 1). As part of our extensive electrochemical investigation, we observed that the mono-brominated spirodienone (type-A), electrochemically constructed from the corresponding phenols, was converted into the dihydrobenzopyrans (type-B) under Lewis acid conditions. A direction of the rearrangement was conducted by the bromo-substituent: the B-1-type products were preferred to the B-2 (ref. 2). As an application of this conversion into the corresponding benzopyran derivatives, a synthesis of heliannuol E (1) will be presented.

Heliannuol E 1 was isolated as a component of Helianthus annuus L. cv. SH-222. This sesquiterpene might be related to the allelopathogenic action of cultivar sunflowers (ref. 3). To accomplish the total synthesis of 1, anodic oxidation of 2 was attempted (CCE, anode: glassy carbon beaker, cathode: platinum wire,  $LiClO_4$  as a supporting salt, MeOH, MeCN, acetone or dioxane with/without 60% aq. HClO<sub>4</sub> as a solvent). Although 2a  $(R_1 = R_2 = R_3 = H)$  provided only trace amounts of **3a** with considerable amounts of by-products, **2b** (R<sub>1</sub>= Br, R<sub>2</sub>= R<sub>3</sub>= H) produced the desired spiro derivative 3b in 50% yield. Based on this observation, appropriate substituents were introduced to the side-chain part of 2 to give 2c (R<sub>1</sub>= Br,  $R_2$ = H,  $R_3$ = CMe<sub>2</sub>OH) and 2d ( $R_1$ = Br,  $R_2$ = (CH<sub>2</sub>)<sub>2</sub>OAc,  $R_3$ = CMe<sub>2</sub>OH). Consequently, both of the phenols afforded the spiro derivatives 3c, 3d, and their structures were unambiguously confirmed by spectroscopic techniques. The following treatment of 3 with  $BF_3OEt_2$  effected the rearrangement to the dihydrobenzopyrans 4 and 5 in good selectivity. As expected, the desired 4 was preferentially produced (ca. 5:1), probably owing to a combination of the opposite inductive effects of the bromine and methyl substituents.

Details of our research progress to the target molecule will be discussed.

## References

- Yamamura, S.; Nishiyama, S. In *Studies in Natural Products Chemistry*, Vol. 10; Atta-ur-Rahman Ed.; Elsevier Science publishers: Amsterdam, 1992; p 629 -669. Yamamura, S.; Nishiyama, S. *J. Syn. Org. Chem., Jpn.* **1997**, *55*, 1029 - 1039. Yamamura, S.; Nishiyama, S. *Synlett* **2002**, 533 - 543.
- Mori, K.; Yamamura, S.; Nishiyama, S. *Tetrahedron* 2001, *57*, 5527 - 5532. Mori, K.; Yamamura, S.; Nishiyama, S. *Tetrahedron* 2001, *57*, 5533 - 5542.
- Macías, F. A.; Varela, R. M.; Torres, A.; Molinillo, J. M. G. *Tetrahedron Lett.* **1999**, *40*, 4725 - 4728

(isolation, structure). Sato, K.; Yoshimura, T.; Shindo, M.; Shishido, K. *J. Org. Chem.* **2001**, *66*, 309 - 314 (synthesis).

