

Influence of Addition of VGCF on LiFePO₄ Cathode Materials in Li-ion Batteries

Chin-Cheng Chen, Mao-Huang Liu, and Jin-Ming Chen^Z

Materials Research Laboratories,
Industrial Technology Research Institute
Hsinchu, Taiwan 31040, R.O.C.

^ZE-mail: JinMingChen@itri.org.tw

1. Abstract

Electrochemical properties of olivine-type LiFePO₄ composite cathode materials prepared by adding VGCF (Vapor-Grown Carbon Fiber) were investigated in this study. The LiFePO₄ and VGCF/LiFePO₄ samples were characterized by X-ray diffraction (XRD) and scanning electron microscope (SEM). Results of experiments indicated that VGCF/LiFePO₄ cathode exhibited good rate capability and high energy density. By adding VGCF content of 10%, the first discharge capacity of 139mAh/g can be achieved under an average discharge voltage of 3.4V at room temperature. This attributed that the addition of VGCF enhanced high electronic conductivity.

2. Experimental

The VGCF/LiFePO₄ composites were prepared by the solid-state reaction of Li₂CO₃ (>99%, J. T. Baker), Fe(II)C₂O₄·H₂O (98.5%, SHOWA), (NH₄)₂HPO₄ (98.5%, SHOWA), and VGCF (VGCF-H, SHOWA). These precursors were dispersed into ball milling then thoroughly mixed. The LiFePO₄ was synthesized in a purified Ar atmosphere to prevent the formation of Fe³⁺ compounds as impurities. The mixture was first decomposed at 350°C for 10hr to expel the regroup in Ar gas, then sintered for 10hr at temperature 700°C in Ar gas.

The XRD (Philips 1700) with Cu Kα radiation was used to identify the crystal structure. The powder morphologies using the SEM were observed (LEO 1530).

Electrochemical characterization of the LiFePO₄ cathode materials was carried out by using a coin-type cells (size 2032). All electrode preparation and sealing were performed in a dry air atmosphere. These cathodes were made by mixing the VGCF containing LiFePO₄ powder with polyvinylidene difluoride (PVDF) binder in a weight ratio of 91:9. The slurry was prepared by blending these precursors with N-methylpyrrolidone (NMP). The solvent was evaporated at 100°C under vacuum, then the dried cathode mixture was compressed into pellets on an aluminum foil. The electrolyte was a solution of 1M LiPF₆ in a 1:1 mixture of ethylene carbonate (EC) and dimethyl carbonate (DMC) (SAMSUNG Chemicals). A lithium metal foil was used as the anode and a microporous polypropylene sheet was used as the separator.

3. Results and Discussion

The LiFePO₄ and VGCF/LiFePO₄ powders with 5-32 μm in particle size are shown in Fig. 1(a) and Fig. 1(b), respectively. The LiFePO₄ powder is olivine-type structure, as shown in the XRD pattern (Fig. 2). The typical charge-discharge curves of LiFePO₄ and VGCF/LiFePO₄ were illustrate in Fig. 3. The result shows that the addition of 10% VGCF to LiFePO₄ cathode shows a higherspecific capacity of 139mAh/g.

References

1. A. K. Padhi, K. S. Najundaswamy, and J. B. Goodenough, *J. Electrochem. Soc.*, **144**, 1188 (1997).

2. A. S. Andersson, J. O. Thomas, B. Kalska, and L. Häggström, *Electrochem. Solid-State Lett.*, **3**, 66 (2000).
3. Z. H. Chen, and J. R. Dahn, *J. Electrochem. Soc.*, **149**, A1184 (2002).
4. A. Yamada, S. C. Chung, and K. Hinokuma, *J. Electrochem. Soc.*, **148**, A224 (2001).
5. M. Takahashi, S. Tobishima, K. Takei, and Y. Sakurai, *J. Power Sources*, **97-98**, 508 (2001).

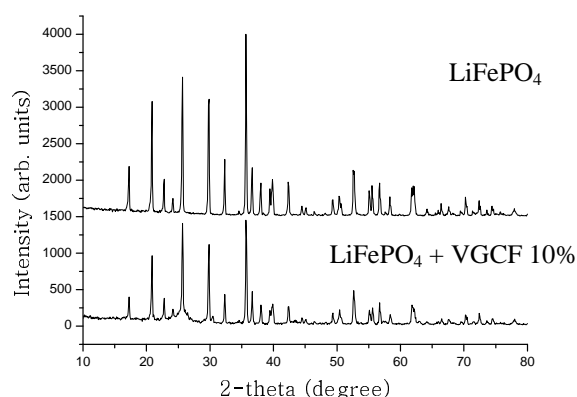
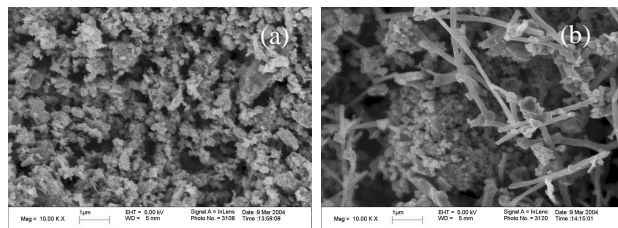


Fig. 1. SEM micrograph of LiFePO₄ powder (a) LiFePO₄ (b) LiFePO₄ + VGCF 10%.

Fig. 2. XRD patterns of the LiFePO₄ and VGCF/LiFePO₄ powder.

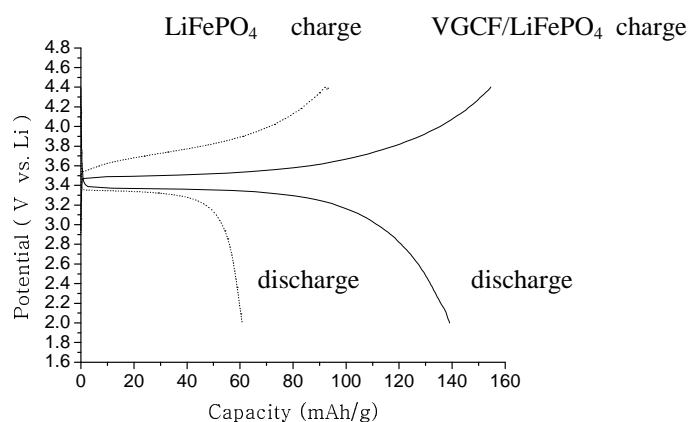


Fig. 3. Typical charge-discharge curves for LiFePO₄ and VGCF/LiFePO₄ cathodes at a current density of 0.14 mA/cm².