

High-Throughput Methods for the Investigation of Photoelectrochemical Hydrogen Production from $Zn_{1-x}Co_xO$ Thin Films

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The solar driven photolysis of water has long been identified as one of the “holy grails” of chemistry because of its potential impact as a clean, renewable, method of fuel production. It is well known that the photoelectrochemical activity of ZnO is limited by two primary factors: (1) poor solar spectrum photon absorption (due to a wide bandgap of 3.2 eV), and (2) photocorrosion in electrolytes [1]. The combinatorial chemistry methodology of rapid synthesis and high-throughput characterization has expedited the discovery of improved solid-state materials for a variety of applications. We have applied both traditional and combinatorial methods to improve the opto-electrochemical properties of hydrogen producing photocatalytic metal oxides. Previous researchers have identified cobalt-doping as a means to improve visible photon absorption in ZnO, however, only a handful of cobalt-loadings were prepared and studied [2-5]. Herein, we present results of a 120-member $Zn_{1-x}Co_xO$ ($0 < x < 0.068$) thin film library that spans 27 different cobalt concentrations with at least 4 replicates each, Fig. 1.

$Zn_{1-x}Co_xO$ samples (200 nm thick) were electrochemically deposited using a rapid serial synthesis system, Fig. 2. Zn:Co stoichiometry in the film was controlled by ratios of $ZnCl_2$ and $Co(NO_3)_2$ in the independent electrochemical deposition cells. This synthetic route followed the ZnO preparation proposed by Gal. et. al. [6]. XPS was used to determine Zn:Co stoichiometries of the doped films, and it was found that the cobalt was predominantly Co^{2+} . XRD confirmed the wurtzite structure for all samples. UV-Vis spectroscopy showed increase visible absorption of the cobalt-doped samples through two means: (1) characteristic *d-d* transitions from the Co^{2+} ion, and (2) decreased bandgap.

A high-throughput photoelectrochemical screening system was developed to characterize the photocatalytic activity of the samples. Fig. 3 shows the zero-bias photocurrent of the samples in 0.2M KNO_3 (aq.). A ~ 1 W/cm² Xe lamp was used for illumination. Under UV-Vis illumination, the photocurrent of the samples asymptotically decreased with increasing cobalt. Under Vis-only illumination, however, photocurrent initially decreases but then rises rapidly to a maximum at Co = 4.4% with a value nearly four times greater than that of pure ZnO. High-throughput photoelectrochemical screening showed greatly increased photoactivity under applied bias, and it was confirmed by action spectra that the Co^{2+} photon absorptions and decreased bandgap of cobalt-doped films observed in UV-Vis spectroscopy gave rise to the increased photocurrent. Flat-band potential measurements indicated a negligible change in the conduction band edge for cobalt-doped samples compared to ZnO, implying that the bandgap shift must arise from an increase in valence band energy.

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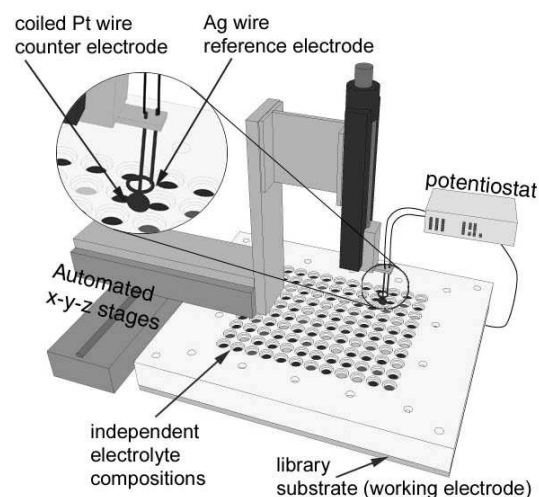


Figure 1. Automated electrochemical synthesis of 120 samples using scanning counter and reference electrodes.

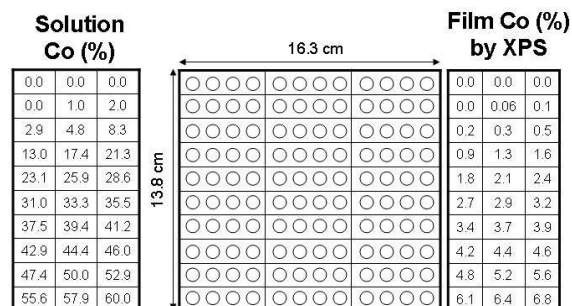


Figure 2. Library design for 120 $Zn_{1-x}Co_xO$ films ranging from $0 < x < 0.068$.

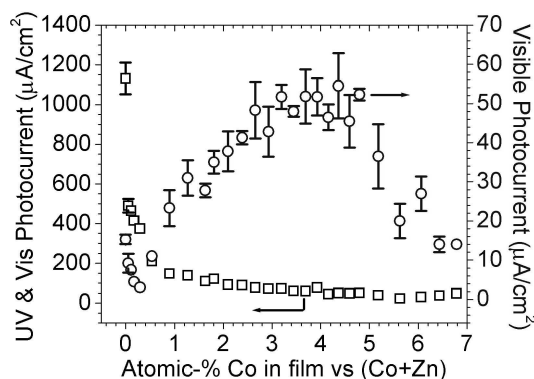


Figure 3. Photocurrent results in 0.2M KNO_3 . Left axis reflects full Xe lamp illumination while the right axis exhibits results with a UV cut-off filter in place.