ELECROCHEMICAL REDUCTION OF PLUTONIUM (III) IN MOLTEN NaCl-KCl AT 740-750C

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High purity plutonium is prepared by electrorefining impure plutonium metal in an equimolar NaCl-KCl melt at 740-750C. The electrochemical reduction occurs via a threeelectron reversible process at a tungsten electrode with the product metal being deposited at the annulus of a concentric magnesium oxide crucible. In addition to seving as a means of preparing high purity plutonium metal, the process is used to recycle plutonium metal scrap. Principals of operation, cell components, materials of construction, corrosion, and hardware improvements will be discussed.