New Hydrophobic Ionic Liquids (Molten Salts) with Highly Fluorinated Anions. Synthesis and Properties.

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Commonly used ionic liquids with the hexafluorophosphate, PF_6 anion (hydrophobic) are hydrolytically unstable, especially by heating [1]. Recently Merck KGaA (Darmstadt, Germany) has developed convenient synthesis of hydrophobic ionic liquids with tris(perfluoroalkyl)trifluorophospate (FAP) - anion [2,3], as replacement for PF_6 .

 $\begin{array}{c|c} Preparation \ of \ the \ ionic \ liquids \ with \ FAP-anion \\ is \ based \ on \ the \ use \ of \\ tris(perfluoroalkyl)trifluorophosphoric \ acid \ of \ type \ (II) \ or \\ its \ alkali-metal \ salts. \end{array}$

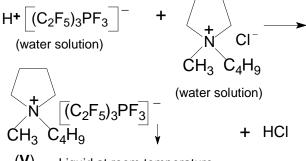
$$\begin{array}{ccc} (C_2F_5)_3PF_2 + HF + 5H_2O \longrightarrow H^+ \left[(C_2F_5)_3PF_3 \right]^- \cdot 5H_2O \\ (I) & (II) \end{array}$$

Phosphorane (I) can be produced by well established industrial method - electrochemical fluorination (ECF) in anhydrous HF.

The aqueous solution of acid (**II**) is stable at room temperature for a long time and can be used as a convenient starting material for the preparation of different salts containing the tris(perfluoroalkyl)trifluorophosphate-anion (FAP-anion) for application as a new ionic liquids and conducting salts, for example:

	$[(C_6H_{13})_3(C_{14}H_{29})P]^+$
CH ₃ -N+N-C ₆ H ₁₃	$[(C_2F_5)_3PF_3]^-$
(III) [(C ₂ F ₅) ₃ PF ₃] ⁻	(IV)

The synthesis of room temperature molten salts with FAP-anion can be carried out in water.



(V) Liquid at room temperature

HMIM FAP (**III**) can be obtained with a very low content of chloride and residual water (10-15 ppm). HMIM FAP (**III**) is a hydrophobic room temperature ionic liquid, which possesses high hydrolytic stability (no HF formation after 5 hours boiling in water) and large electrochemical window (more than 5.5 V). Viscosity (mm²/s) of HMIM FAP (**III**) is: 74.3 (20°C), 29.8 (40°C), 14.8 (60°C), 8.4 (80°C).

 $Tetrafluoroborate, BF_4-anion \ can \ be \ modified \ in the same way. The introduction of perfluoroalkyl-groups to boron makes the room temperature molten salts with perfluoroalkyl-$

temperature molten saits with perfluoroalkyltrifluoroborate-anion hydrophobic and increase its electrochemical stability. Recently Merck KGaA (Darmstadt, Germany) has developed new method of synthesis of the ionic liquids with perfluoroalkyltrifluoroborate-anion [4].

The electrochemical properties (conductivity and electrochemical stability) of new ionic liquids (room temperature molten salts) will be discussed.

References.

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