

## SMART CARBON DIOXIDE GAS SENSOR BASED ON SOLID ELECTROLYTES

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### Introduction

Recently, the CO<sub>2</sub> monitoring has been becoming an urgent issue not only for the suppression of the “greenhouse effect”, but also for the amenity in facilities. Among various sensing methods, all solid state type sensor is one of promising tools for in situ monitoring. From our previous researches on the CO<sub>2</sub> gas sensors based on solid electrolytes (Figure 1), rapid, continuous, and reproducible sensing performances have been demonstrated.<sup>1-4</sup>

In this study, the miniaturized sensor cell of the size in 2mm, was fabricated with the thin Pt film heater, and its practical sensing performance was investigated.

### Experimental

Figure 2 presents a schematic illustration of the miniaturized CO<sub>2</sub> gas sensor. The sensor consists of the detecting La<sub>2</sub>O<sub>2</sub>SO<sub>4</sub>-Li<sub>2</sub>CO<sub>3</sub> auxiliary electrode with Au sensing electrode, cationic/anionic conductor tip, Pt reference electrode and alumina substrate with a thin Pt film heater. After co-pressing the individual solid powder of the cationic/anionic conductor, the cationic/anionic conductor tip was sintered and then cut into pieces of 0.5mm<sup>2</sup>. The tip was fixed to the Al<sub>2</sub>O<sub>3</sub> substrate with a binder. The detecting and the reference electrodes were prepared by using the La<sub>2</sub>O<sub>2</sub>SO<sub>4</sub>-Li<sub>2</sub>CO<sub>3</sub> powder and the Pt pastes on each side of the tip, respectively. The operation temperature was controlled at 500 with electric power consumption of 0.5W and the CO<sub>2</sub> sensor output (EMF) was monitored in the CO<sub>2</sub> concentration range between 500ppm and 3%.

### Results and Discussion

Figure 3 depicts the variation of the sensor output (EMF) with the CO<sub>2</sub> gas concentration. A linear relationship was clearly observed between the sensor output and log(PCO<sub>2</sub>). The slope obtained from the EMF output vs. log(PCO<sub>2</sub>) relationship is n=1.92, which well coincides with that from the theoretical Nernst equation (solid line in Fig. 3). The above result explicitly indicates the fact that the CO<sub>2</sub> gas concentration can be quantitatively determined only by measuring the sensor output.

Figure 4 shows the long term stability of the present CO<sub>2</sub> sensor. A high stability was ensured after ca. 100 days, suggesting a satisfactory performance for in situ monitoring.

### References

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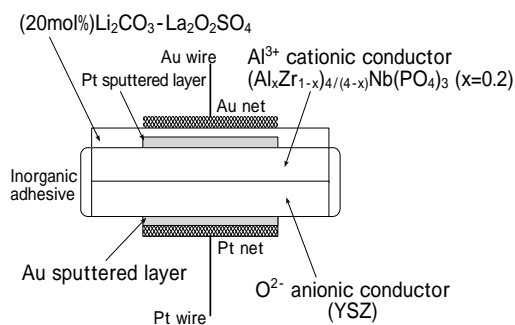


Fig. 1. The cross-sectional view of the sensor element based on Al<sup>3+</sup> and O<sup>2-</sup> ion conducting solids with La<sub>2</sub>O<sub>2</sub>SO<sub>4</sub>-Li<sub>2</sub>CO<sub>3</sub> auxiliary electrode.

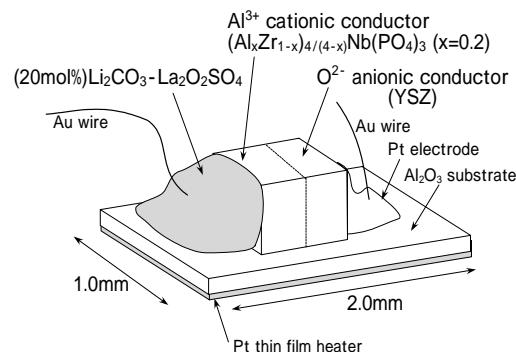


Fig. 2. A schematic illustration of the miniaturized CO<sub>2</sub> sensor element with thin Pt film heater.

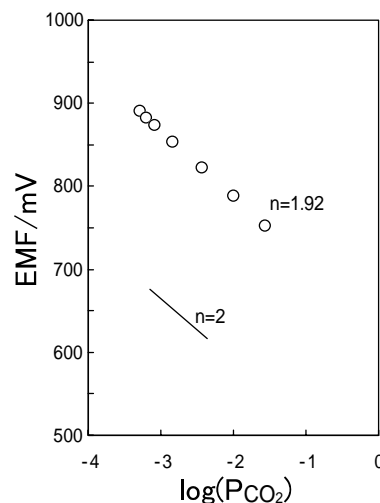


Fig. 3. The relationship between the sensor output(EMF) and the logarithm of the CO<sub>2</sub> concentration at 500 .

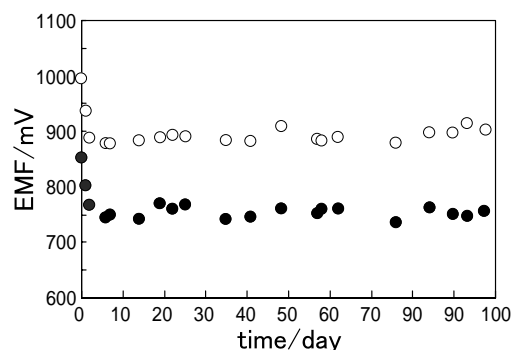


Fig. 4. The long term stability of the sensor output(EMF) for the CO<sub>2</sub> gas concentration of 500ppm( ) and 3% ( ) .