

## Thin Film Mixed Potential NO<sub>x</sub> Sensor Development for Stationary Reciprocating Engine Applications

Eric L. Brosha, Rangachary Mukundan, Roger Lujan, and Fernando H. Garzon

*Los Alamos National Laboratory  
Electrochemical Materials and Devices Group  
Los Alamos, New Mexico 87545*

Advanced reciprocating engines require new sensor technologies to reduce emissions and to improve overall combustion efficiency. Engine feedback control, NO<sub>x</sub> emissions monitoring and control, and sensors to control the regeneration of reduction catalysts are required. Some of the principal goals of the DOE's Advanced Reciprocating Engine System (ARES) program are to improve fuel efficiency and flexibility, and reduce emissions of engine systems used for distributed energy generation applications. The NO<sub>x</sub> emissions target for natural gas-fired reciprocating engines is 0.1 g/bhp-hr (roughly a factor of ten reduction from current levels). A new NO<sub>x</sub> sensing technology is necessary to meet the sensor requirements for low cost, sulfur tolerance, and a long, stable lifetime. The sensors developed over the last decade for lean-burn gasoline engines cannot meet all of these requirements. In our past work, we have developed CO and HC mixed potential sensors based on doped zirconia [1-3] and ceria [4] electrolytes and metal oxide electrodes for automotive on-board diagnostic applications. We are presently engaged in work to determine whether our sensors would be suitable for NO<sub>x</sub> sensing and combustion control in large, natural gas fired reciprocating engines.

Mixed potential sensors based on Mg and Sr-doped lanthanum chromite and Pt electrodes with thin film, YSZ solid electrolyte are studied for NO/NO<sub>2</sub> sensitivity, CO/CH<sub>4</sub>/HC cross sensitivity, and voltage level stability in this work. Thin films of La<sub>0.8</sub>(Sr,Mg)<sub>0.2</sub>CrO<sub>3</sub> and LaCr<sub>0.8</sub>Mg<sub>0.2</sub>O<sub>3</sub> were prepared using RF magnetron sputtering via a mixed phase, fluoride precursor route [5]. Thin films of YSZ electrolyte (< 15μm) were prepared using e<sup>-</sup> beam evaporation. Planar sensor configurations were selected to facilitate the future incorporation of a thin film heater.

Figure 1 is the initial sensor response to NO<sub>2</sub> at 650°C in a base atmosphere of 10.4% O<sub>2</sub>. The device response showed a fast response time with excellent reproducibility in the voltage levels with changing NO<sub>2</sub> concentrations. The temperature was lowered to 600°C with little change in the NO<sub>2</sub> level (small temperature coefficient) as can be seen from Figure 2. Figure 2 shows the cross sensitivity to CO and to a few selected HCs; particular attention was made to the cross sensitivity to CH<sub>4</sub>. The aging effects were also studied for NO<sub>2</sub> (Figure 3) along with NO, CO, C<sub>3</sub>H<sub>6</sub>, C<sub>3</sub>H<sub>8</sub>, and CH<sub>4</sub>. In addition, the PO<sub>2</sub> dependence was studied as well for this sensor.

Initial experiments have also been performed to demonstrate the use of pre-catalysts to enhance selectivity. Multi-electrode devices and the use of pre-catalysts were studied as potential methods to mitigate cross sensitivity to non-methane HCs and CO as well as to control NO/NO<sub>2</sub> selectivity at different temperatures. We will also present in this talk some preliminary work in the exploration of doped fluorite and spinel metal oxide thin films as potential NO<sub>x</sub>-specific working electrodes for mixed potential sensors.

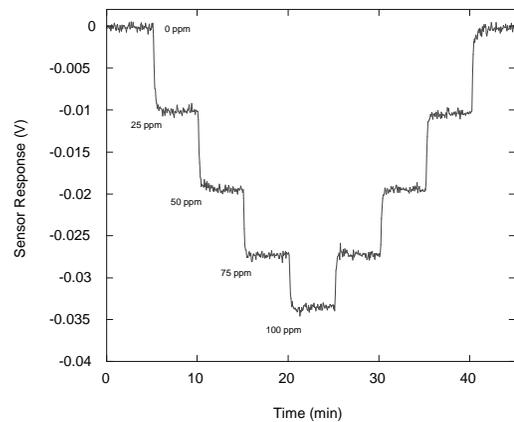


Figure 1. Initial NO<sub>2</sub> response, 0-100 ppm, at 650°C, 10.4 % O<sub>2</sub>/N<sub>2</sub> balance flowing at a base flow rate of 500 SCCM.

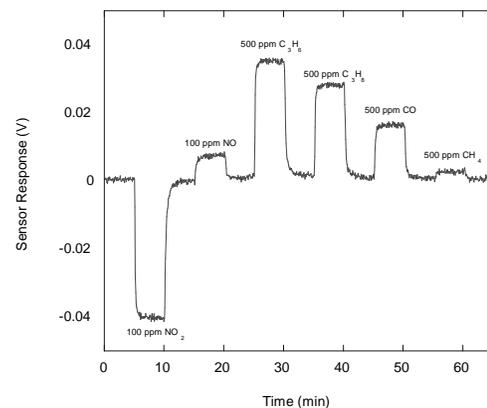


Figure 2: Cross sensitivity studies: sensor response to NO<sub>2</sub>, NO, and selected HC's and CO versus time at 600°C. Base gas is 10.4% O<sub>2</sub> at a flow rate of 500 SCCM.

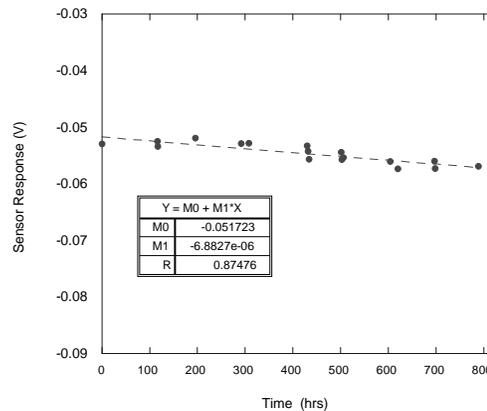


Figure 3: Plot of sensor voltage measured at 100 ppm of NO<sub>2</sub> versus a time period of 800 hours at 600°C in 10.4% O<sub>2</sub>.

### References:

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