

## Photo-enhanced wet oxidation and etching of GaN

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The wide bandgap III-V nitrides have long been recognized as the material of choice for realizing short wavelength emitters, high temperature and power electronics. The chemical stability of III-V nitrides, however, prevents the use of cost-effective wet methods in device processing. In this work, we report the use of photoelectrochemical (PEC) method to achieve wet oxidation, etching, and MOS action on n-type GaN and enhanced photo-response on oxidized InGaN QW.

The PEC set-up was similar to that originally designed by Haisty [1] in which the galvanic cell consisting of a GaN working and a Pt counter electrode immersed in an electrolyte. Activation of the PEC process was initiated by a 254 nm mercury line source at an intensity  $\sim 10 \text{ mW/cm}^2$ . The reaction is accredited to the UV-excited hot carriers at the GaN/electrolyte interface to access the  $\text{H}^+/\text{H}_2$  and  $\text{OH}^-/\text{O}_2$  redox levels and to enhance the oxidative dissolution of GaN. [2]

Shown in Fig. 1 is the PEC response from n-type GaN grown on sapphire. Wet oxidation of GaN is found to take place in  $\text{H}_3\text{PO}_4$  solutions with pH values ranging from 3 to 4, whereas etching can occur at a larger window from  $-1$  to 2 in  $\text{H}_3\text{PO}_4$  and 11 to 15 in KOH solution. At room temperature, the PEC process is found reaction-rate limited and has a peak value of 224 nm/hr at pH = 3.5 for oxidation and 125 nm/min at pH = 0.75 for etching.

Illustrated in Fig. 2 are the SEM micrographs taken on the PEC oxidized GaN with an oxide thickness of (a) 100 nm and (b) 450 nm. The thin-oxidized sample A has a purple-bluish color whereas cracks appear on the thick sample B due to strain relief during the oxide growth. The missing N signal in the Auger spectra of Fig. 2 (c) suggests a complete surface coverage by an oxide layer, whereas the up shifting of the Ga  $2\text{P}_{3/2}$  binding energy of the oxidized GaN to 1119 eV in Fig. 2 (d) represents a typical signal from  $\text{Ga}_2\text{O}_3$ . [3]

The spectral photo-current (PC) response of Fig. 3 reveals an optical enhancement by a factor of (a) 3 and (b) 40% in the oxidized GaN and InGaN MQW, respectively. These observations suggest that the PEC oxide can be used as an efficient surface passivation layer on the GaN-based opto-electronic devices.

In Fig 4 we depict the high frequency (1 MHz) C-V characteristics of a GaN MOS with a 50 nm-thick PEC-grown  $\text{Ga}_2\text{O}_3$  and a size of  $\sim 1.7 \times 10^{-4} \text{ cm}^2$ . The data reveals a sharp transition between the depletion and accumulation regimes, and has a minimum interface trap density estimated to be  $D_{it, \min} < 2 \times 10^{12} \text{ cm}^{-2} \text{ eV}^{-1}$ .

In summary, we report a PEC wet oxidation and etching process on GaN and demonstrate their promising use as the surface passivation layer to the GaN-based opto- and electronic devices.

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Reference:

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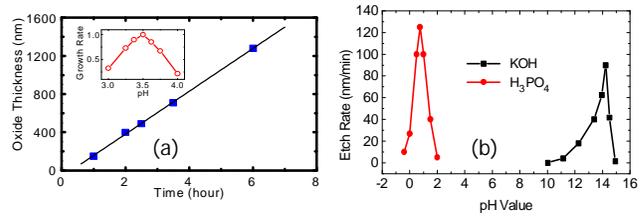


Fig. 1 PEC (a) oxidation, and (b) etch rate of n-GaN

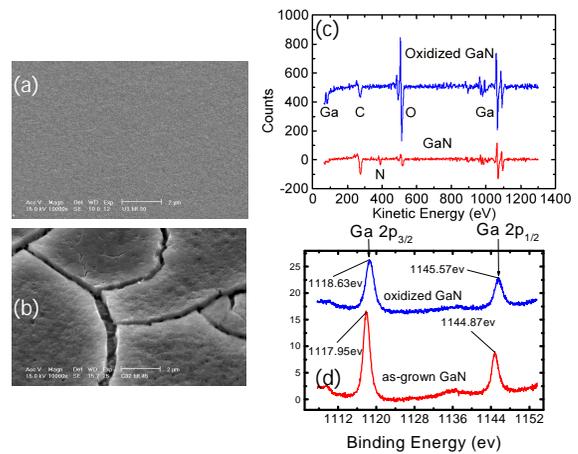


Fig. 2 (a), (b): SEM micrograph, (c) Auger, (d) XPS spectra of PEC-grown  $\text{Ga}_2\text{O}_3$  on GaN.

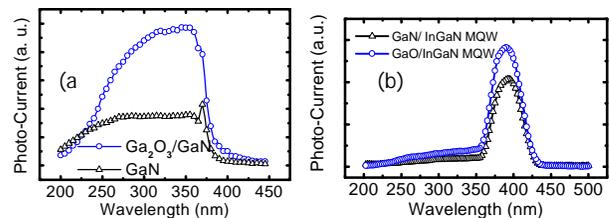


Fig. 3 Oxidation enhanced photo-current response on GaN, and (b) InGaN MQW

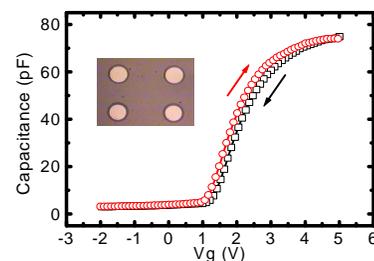


Fig. 4 High frequency (1 MHz) C-V data of  $\text{Ga}_2\text{O}_3/\text{GaN}$