

# DIRECT SONOCHEMICAL PREPARATION OF HIGHLY PHOTOACTIVE MESOPOROUS TITANIUM DIOXIDE WITH A BICRYSTALLINE FRAMEWORK

Li-Zhi Zhang, Jiaguo Yu, Jimmy C. Yu

Department of Chemistry and Environmental Science Programme, The Chinese University of Hong Kong, Shatin, New Territories, Hong Kong, China

Mesoporous  $\text{TiO}_2$  with a bicrystalline (anatase and brookite) framework was synthesized directly without thermal treatment under high intensity ultrasound irradiation in the absence (SM-1) and presence (SM-2) of triblock copolymer (Fig. 1). In the absence of triblock copolymer, the mesoporous  $\text{TiO}_2$  was formed by the agglomeration of monodispersed  $\text{TiO}_2$  sol particles (Fig. 2). In the presence of a triblock copolymer as a structure-directing agent, the crystalline size, pore size (Fig. 3), and brookite content were all increased. XRD, TEM, nitrogen adsorption, TGA/DTA and FTIR were used for the characterization of mesoporous  $\text{TiO}_2$ . Ultrasound irradiation and the use of triblock copolymer seem to be beneficial to the formation of brookite phase. Both as-prepared mesoporous  $\text{TiO}_2$  samples show better activities than commercial photocatalyst Degussa P25 in degradation of acetone in air. The activities of mesoporous  $\text{TiO}_2$  after calcination increase because of better crystallization. The best mesoporous  $\text{TiO}_2$  is 80% more active than P25. The high activities of the mesoporous  $\text{TiO}_2$  with a bicrystalline framework can be attributed to the combined effect of the presence of brookite, high surface area and interconnection of mesopores.

References:

1. J. C. Yu, J. Lin, R. W. M. Kwok, *J. Phys. Chem. B.* 1998, **102**, 5094.
2. D. M. Antonelli, Y. J. Ying, *Angew. Chem. Int. Ed. Engl.* 1995, **34**, 2014.
3. Y. Wang, X. Tang, L. Yin, W. Huang, Y. R. Hacoheh, A. Gedanken, *Adv. Mater.* 2000, **12**, 1183.
4. J. C. Yu, J. G. Yu, W. K. Ho, L. Z. Zhang, *Chem. Commun.* 2001, 1942.

5. K. S. Suslick, G. J. Price, *Ann. Rev. Mater. Sci.* 1999, **29**, 295.
6. J. C. Yu, L. Z. Zhang, J. G. Yu, *New J. Chem.*, in press.

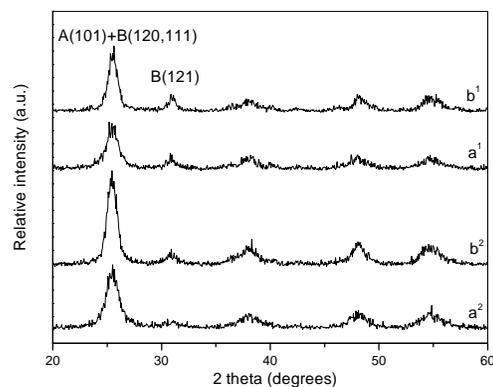


Fig. 1. XRD patterns of SM-1 and SM-2, A denotes anatase, B denotes brookite,  $a^1$  denotes the as-prepared SM-1,  $b^1$  denotes the calcined SM-1,  $a^2$  denotes the as-prepared SM-2,  $b^2$  denotes the calcined SM-2.

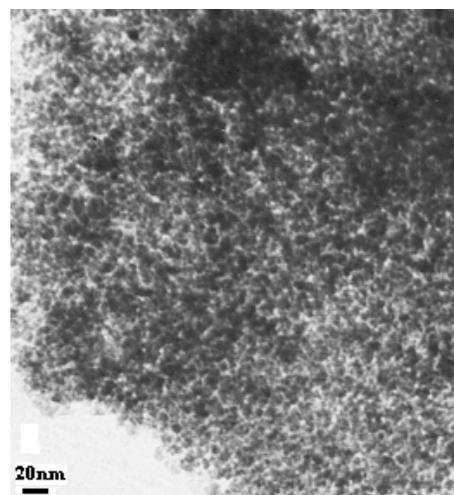


Fig. 2. TEM image of the as-prepared SM-1.

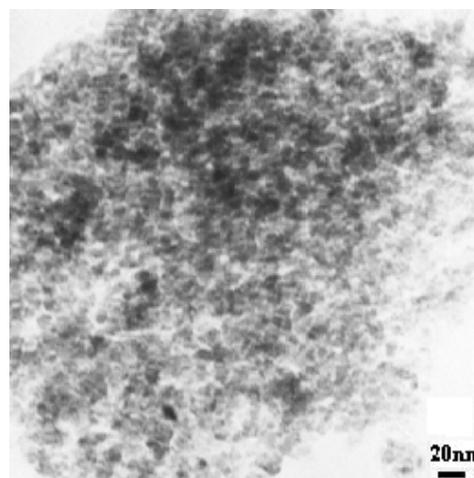


Fig. 3. TEM image of the as-prepared SM-2.