

Electronic structure of LiCoO₂ and LiNiO₂ bulk materials
and thin films prepared by magnetron sputtering

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LiCoO₂ and LiNiO₂ are most promising materials for
applications as cathode materials in thin film battery
devices.

In this work bulk powder samples and thin films
deposited by magnetron sputtering have been used to
study the electronic structure of the stoichiometric
compounds and during Li-deintercalation and
intercalation in order to elucidate its relation to
electrochemical parameters like battery voltages.

Commercial LiCoO₂ and LiNiO₂ powders have been
pressed to pellets to improve handling in the vacuum
system. Different vacuum preparation techniques like
Ar⁺-ion etching and annealing in different oxygen partial
pressures have been used to obtain clean and
stoichiometric sample surfaces. The effects of the
different preparation steps on surface contaminations and
oxygen and lithium content in the samples has been
investigated systematically.

LiCoO₂ thin films have been prepared by magnetron
sputtering. First results on the effect of different
sputtering parameters like argon and oxygen partial
pressure and sputtering voltages and frequencies will be
presented.

The crystal structures and sample morphologies have
been studied by X-ray diffraction (XRD), scanning
electron (SEM) and atomic force microscopy (AFM).

X-ray induced photoelectron spectroscopy (XPS) has
been used to determine chemical composition and
stoichiometry of the samples and the oxidation states of
the different elements in the compound. UV-light induced
photoelectron spectroscopy (UPS) and resonant
photoemission (ResPES) has been used to study the
electronic structure. While UPS reveals information on
work function and band bending, ResPES allows the
assignment of valence band emissions to O2p-, Co3d- or
Ni3d-like states. Thus, the partial valence band density of
states (PVBDOS) can be derived.

First results on electrochemical measurements performed
in a battery setup for the powder samples and for the thin
films will be presented. Ex-situ cycling of the cells
influences the chemical and electronic properties of the
cathode materials studied by XPS.