

# Protonic Conductivity Measurements of CsHSO<sub>4</sub>/SiO<sub>2</sub> Composite Electrolyte for Intermediate Temperature Fuel Cell

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## 1. Introduction

Protonic conductivity measurements in pure CsHSO<sub>4</sub> and CsHSO<sub>4</sub>/SiO<sub>2</sub> systems were carried out in the temperature range of 350 K to 460 K, aiming at developing a solid inorganic electrolyte for intermediate temperature fuel cell. We discuss the influence of heterogeneous SiO<sub>2</sub>-doping into CsHSO<sub>4</sub> salt on the protonic conductivity as a function of mol ratio of SiO<sub>2</sub> by AC impedance spectroscopic measurements.

## 2. Experimental

CsHSO<sub>4</sub>/SiO<sub>2</sub> composites were prepared by grinding in an agate mortar for 30 minutes, and then pressed at 3t/cm<sup>2</sup> and finally calcined at 470 K for 1 h in an air. SiO<sub>2</sub> with meso-pores (particle size: 1.3μm, mesopore diameter: 22 nm, specific surface area: 300 m<sup>2</sup>g<sup>-1</sup>) was used in the present experiments. The conductivity measurements were carried out with Ag-electrodes in Ar atmosphere by Hewlett Packard 4192A impedance analyzer over frequency range of 10Hz-10MHz.

## 3. Results and Discussions

### 3-1 Conductivity of (1-x)CsHSO<sub>4</sub>/xSiO<sub>2</sub> composite

The temperature dependencies of conductivity of CsHSO<sub>4</sub>/SiO<sub>2</sub> were presented in Fig.1. The mol ratio of SiO<sub>2</sub>, x, of (1-x)CsHSO<sub>4</sub>/xSiO<sub>2</sub> composites was varied from 0 to 0.65. The conductivities in both the pure and composites systems abruptly changed at around 410 K from low temperature phase (350-390 K) to high temperature phase (420-460 K). In pure CsHSO<sub>4</sub>, the conductivity-increase of about 3 magnitudes was observed at around 410 K, which was accompanied by a structural phase transition (monoclinic → tetragonal). In CsHSO<sub>4</sub>/SiO<sub>2</sub> composites systems, at any mol ratio of SiO<sub>2</sub>, the conductivity at the low temperature phase increased about one magnitude in comparison with pure CsHSO<sub>4</sub>. The maximum conductivity at the high temperature phase was observed at x = 0.15, while the conductivity at the high temperature phase became smaller with an increase of SiO<sub>2</sub> mol fraction.

### 3-2 Impedance spectra

The impedance plots at 403 K with various mol ratios were shown in Fig.2. While the impedance of the pure CsHSO<sub>4</sub> and the composites with low mol fraction of SiO<sub>2</sub> (x = 0.15) showed one semicircle, two semicircles appeared in the composites with high mol ratio of SiO<sub>2</sub> (x = 0.45, 0.60). Thus, we tried to separate and assign the impedance spectra, assuming an equivalent circuit consisted of a series circuit of two parallel circuits of a resistance and a capacitance. As seen in the Fig.2, the high frequency semicircles became smaller with the increase of SiO<sub>2</sub> additives. On the contrary, low frequency semicircles became bigger. It was considered that this trade-off relation for resistances of high and low frequency semicircles resulted in the maximum conductivity at a relevant mol ratio of SiO<sub>2</sub>. The values of capacitance of high frequency semicircles are similar to that of pure CsHSO<sub>4</sub> (5.2 × 10<sup>-10</sup>F), and those of the low frequency semicircles are larger by 1-2 orders, compared with pure CsHSO<sub>4</sub>. Therefore, it is considered that high frequency semicircles are derived from the proton conduction of bulk and inter-

face transfer between CsHSO<sub>4</sub> and SiO<sub>2</sub>. This interface phase probably enhances the protonic conductivity. Low frequency semicircles are associated with the slow proton conduction at the boundary inhibited by the SiO<sub>2</sub> particles. Thus, by controlling the dispersed structure of the SiO<sub>2</sub> particles, protonic conductivity can be improved. We consider that further investigations are necessary with the samples prepared with a variety of SiO<sub>2</sub>-characteristics, mixing methods, and structures in CsHSO<sub>4</sub>/SiO<sub>2</sub> systems.

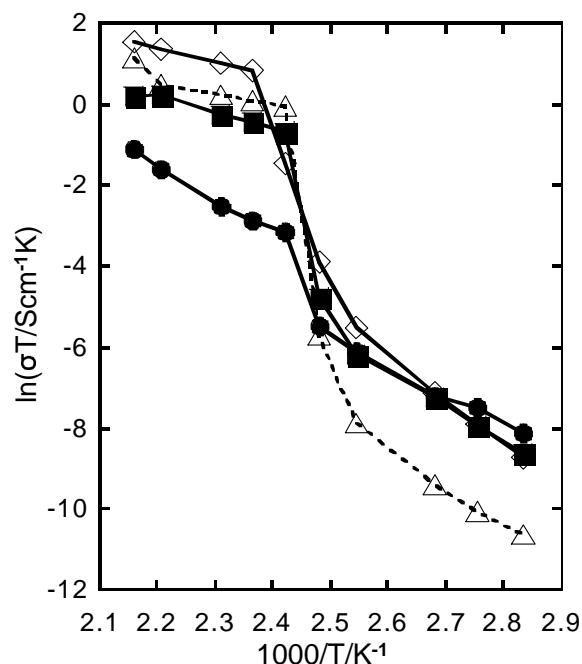


Fig.1 Temperature dependences of the conductivity of (1-x)CsHSO<sub>4</sub>/xSiO<sub>2</sub> composites with different SiO<sub>2</sub> contents and pure CsHSO<sub>4</sub> on cooling process.  $\triangle$  : Pure CsHSO<sub>4</sub>  $\diamond$  : x = 0.15  $\blacksquare$  : x = 0.40  $\bullet$  : x = 0.65

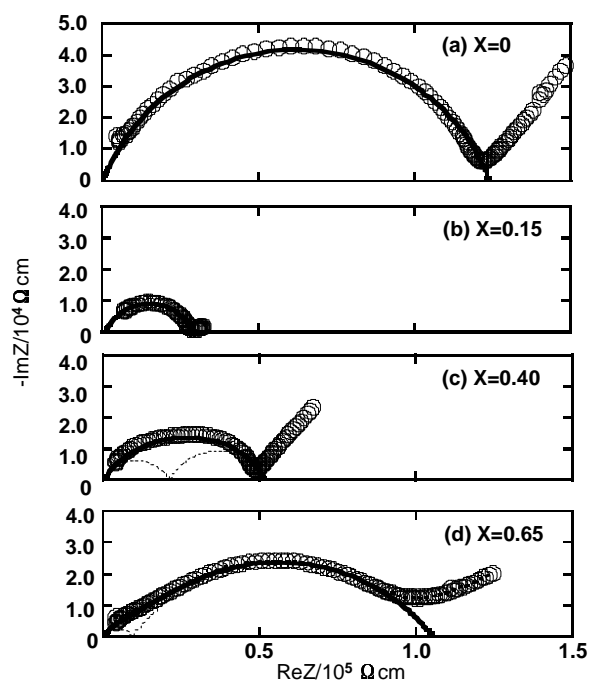


Fig.2 Typical impedance plots of (1-x)CsHSO<sub>4</sub>/xSiO<sub>2</sub> composites at 403 K in Ar atmosphere. Open circle: experimental value; Solid line: fitting curve; Dotted line: deconvoluted semicircles of the fitting curve (a) Pure CsHSO<sub>4</sub> (b) x = 0.15 (c) x = 0.40 (d) x = 0.65