

## Formation Process of Reactively Sputtered MoO<sub>3</sub> Thin Films and Their Optical Properties

Yoshio Abe<sup>1</sup>, Hiroaki Imamura<sup>1</sup>, Eisuke Washizu<sup>1</sup>,  
Katsutaka Sasaki<sup>1</sup> and Toshiaki Shouzu<sup>2</sup>  
<sup>1</sup>Kitami Institute of Technology  
165 Koen-cho, Kitami 090-8507, Japan  
<sup>2</sup>Olympus Optical Co.,Ltd.  
2-3 Kuboyama-cho, Hachioji 192-8512, Japan

Electrochromic MoO<sub>3</sub> thin films are thought to be useful as a neutral light adjustment element for charge coupled device (CCD) and complementary metal-oxide semiconductor (CMOS) imaging device. Reactive sputtering has been commonly used for preparing oxide thin films, however, formation process of reactively sputtered MoO<sub>3</sub> films has not been studied in detail. In this paper the effect of oxygen flow ratio on the formation of MoO<sub>3</sub> films is studied

An RF magnetron sputtering system was used for sample preparation. A 2 inch disk of Mo target (99.9% purity) was sputtered in an Ar and O<sub>2</sub> mixed gas, and Mo and MoO<sub>3</sub> films were deposited on quartz glass substrates. O<sub>2</sub> flow ratio was varied from 0% to 100%. RF power, total gas flow rate and sputtering gas pressure were kept constant at 50 W, 5 cc/min and 50 mTorr, respectively. Plasma emission spectra were monitored by a CCD detector. Optical properties of the MoO<sub>3</sub> films were characterized by spectrophotometry and ellipsometry.

Transmittance of the deposited films is plotted as a function of O<sub>2</sub> flow ratio in Fig. 1. The color of the films deposited at O<sub>2</sub> flow ratios of 0-10% is metallic silver, however, it becomes transparent and transmittance more than 70% is obtained above O<sub>2</sub> flow ratios of 15%. The transparent films are thought to be MoO<sub>3</sub> because refractive index of 1.8-2.1, which is close to that reported for MoO<sub>3</sub> films, was obtained. The deposition rate of the films decreases abruptly at an O<sub>2</sub> flow ratio of 15% as shown in Fig. 2. The decrease of the deposition rate is thought to be caused by the target mode change from metal mode to oxide mode. Figure 3 shows plasma emission intensity of oxygen atoms during the sputter-deposition as a function of O<sub>2</sub> flow ratio. The emission intensity of oxygen atoms is very low at O<sub>2</sub> flow ratios of 0-10% and it begins to increase above 15%. The amount of Mo atoms sputtered from the target was estimated to be  $2.7 \times 10^{-5}$  mol/min in the metal mode by the mass of the deposited Mo film, and the amount of supplied O<sub>2</sub> molecules is calculated to be  $3.35 \times 10^{-5}$  mol/min at an O<sub>2</sub> flow ratio of 15%. Therefore, the ratio of the number of supplied Mo atoms to that of oxygen atoms becomes 1:2.5, which is close to the stoichiometric value of 1:3 for MoO<sub>3</sub>, at the critical O<sub>2</sub> flow ratio of 15%.

From these results the formation process of Mo and MoO<sub>3</sub> films is thought to be as follows. 1) The introduced O<sub>2</sub> molecules are gettered by the Mo atoms deposited on the sputtering chamber wall and Mo films are formed at O<sub>2</sub> flow ratios of 0-10%. 2) Above an O<sub>2</sub> flow ratio of 15%, oxygen density in the plasma begins to increase because the amount of introduced oxygen is more than that necessary to the formation of MoO<sub>3</sub>. Due to the oxygen plasma the surface of the Mo target begins to be oxidized and the deposition rate decreases abruptly. As the results MoO<sub>3</sub> films are formed.

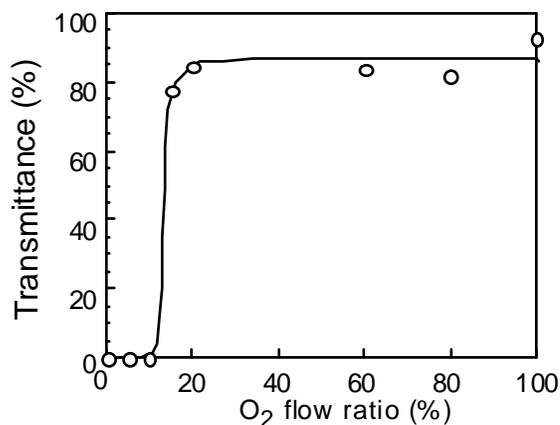


Fig. 1. Transmittance of the films at a wavelength of 600 nm as a function of O<sub>2</sub> flow ratio.

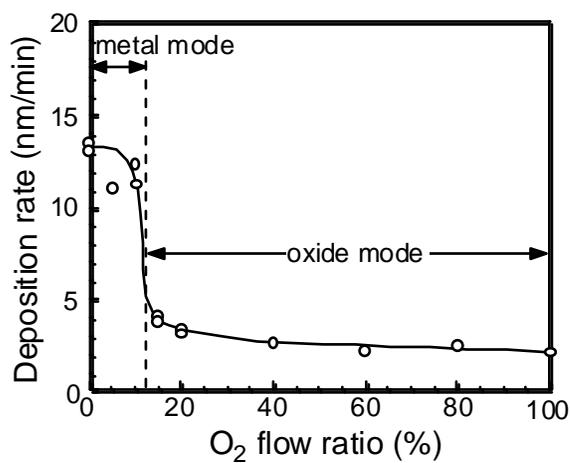


Fig. 2. Deposition rate of the films as a function of O<sub>2</sub> flow ratio.

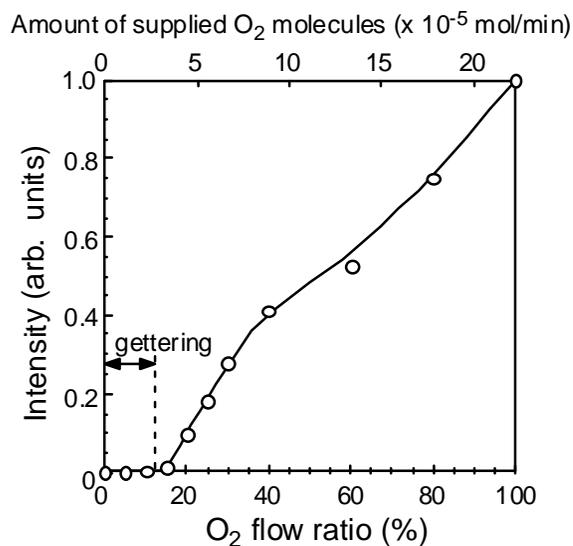


Fig. 3. Plasma emission intensity of oxygen atoms (777 nm) as a function of O<sub>2</sub> flow ratio.