## Electrochromic Effect in Amorphous α-WO<sub>3</sub> as a Outcome of the Occupation of the Bottom of Conduction Band by Electrons

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Electrochromic films of amorphous  $\alpha$ -WO<sub>3</sub> produced by the electrochemical deposition of tungsten-containing compounds were studied with simultaneous determination, during deposition, of current values of thickness and optical refractive index by an interferometric method. The disorder of the structure of these films as numerous terminal double W=O bonds instead of W-O-W bridge bonds is caused by the presence in them of fine pores filled with water molecules (water content up to 15 %). The potential of films was cycled in 1M H<sub>2</sub>SO<sub>4</sub> over a range of +0.4 V to -0.4 V (SCE). Two "waves" were distinguished in cathode region ; reversible coloration of films with light absorption peak, whose position depended on potential and varied from 1.3  $\mu m$  to 0.5  $\mu m$ , corresponded to these waves. The fundamental absorption edge  $\varepsilon_{\alpha}$  of WO<sub>3</sub> shifted by 0.5 V, which corresponded to a change of 0.6 V in the quasi-equilibrium potential E0 of WO3. A shift in E<sub>0</sub> which is observed in this case is caused by a change in the electrochemical potential of electrons  $\varepsilon_{f}$ .

The comparable changes in  $E_0$ ,  $\varepsilon_{\alpha}$  and  $\varepsilon_f$  indicate that the injected electrons occupy sequentially the vacant energy levels of the bottom of conduction band. This lead to an increase on several orders of magnitude in film conductivity and to decrease in refractive index from n=1.68 up to n=1.45 due to a high electron concentration in a film. In this case, despite of degeneration of an electronic subsystem as a result of pairing electrons at d levels, the coloration of films corresponds to electronic intracentre d-d transitions in W. The W ions are in a ligand field, which differs from octahedral one. The higher the degree of ligand field disturbance, the lower the W centre energy and the lower the quantum energy needed for the optical excitation of the corresponding d-d transition.