

**INVESTIGATION OF THE ENERGY TRANSFER
MECHANISMS IN Ho³⁺ - Eu³⁺ CO-DOPED
NANOCRYSTALLINE Y₂O₃**

J. C. Boyer^a, F. Vetrono^a, J. A. Capobianco^a,
A. Speghini^b and M. Bettinelli^b

^aDepartment of Chemistry and Biochemistry, Concordia
University, 1455 de Maisonneuve Blvd. W, Montreal,
Canada

^bDipartimento Scientifico e Tecnologico, Università di
Verona, and INSTM, UdR Verona, Ca' Vignal, Strada Le
Grazie 15, I-37134 Verona, Italy

Over the past several years there has been an ongoing search for nanometer sized powered phosphors with superior performance characteristics over their micrometer counterparts. This research has been stimulated by the fact that significant changes in the optical properties have been observed with decreasing particle size. Much of this recent work has focused on the nanometer equivalent of the common red phosphor europium doped cubic Y₂O₃.

In this paper we report on the spectroscopic studies of a typically utilized red phosphor Y₂O₃: 1 mol% Eu³⁺ co-doped with 1 mol% Ho³⁺. Two nanocrystal samples synthesized through solution combustion and wet chemical methods were examined. Both synthesis routes provide crystallite sizes in the nanometer range but with differing morphologies. A micron sized sample of similar composition was also synthesized through conventional ceramic techniques for comparative purposes.

Under 457.9 nm excitation, an energy transfer process occurs, that affects both the emissive and lifetime properties of excited states. For instance, a quenching of the green (⁵F₄, ⁵S₂ → ⁵I₈) Ho³⁺ emission is noted upon the presence of the Eu³⁺ ion. This is reflected in the decay time of the (⁵F₄, ⁵S₂) levels which exhibits single exponential behavior in the case of a bulk Y₂O₃:Ho³⁺ sample but becomes shorter and non-exponential upon co-doping. As seen in Figure 1, the quenching of the green luminescence is more pronounced in the nanocrystalline samples compared to the conventional micron sample. Excitation with 257.25 nm radiation into the Eu-O charge transfer band results in luminescence almost exclusively from the Eu³⁺ ion. This observations indicate that while the Ho³⁺ → Eu³⁺ energy transfer is efficient at current doping levels the Eu³⁺ → Ho³⁺ energy transfer is not.

The effect of Eu³⁺ presence on the upconversion properties (conversion of near IR to visible light) of the Ho³⁺ ion is also examined. The upconversion efficiency is reduced in the co-doped sample with respect to the singly doped sample due to depopulation of the ⁵I₇ level by an energy transfer process to the ⁷F₆ level of the Eu³⁺ ion.

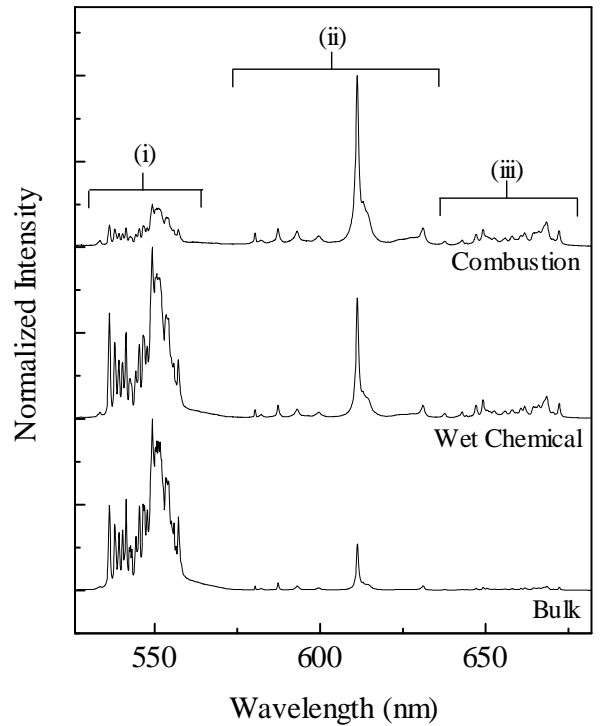


Figure 1: RT luminescence of Y₂O₃:Eu³⁺, Ho³⁺ samples upon excitation at 457.9 nm. (i) ⁵F₄, ⁵S₂ → ⁵I₈ [Ho³⁺] (ii) ⁵D₀ → ⁷F₀, ⁷F₁, ⁷F₂ [Eu³⁺] (iii) ⁵F₅ → ⁵I₈ [Ho³⁺]

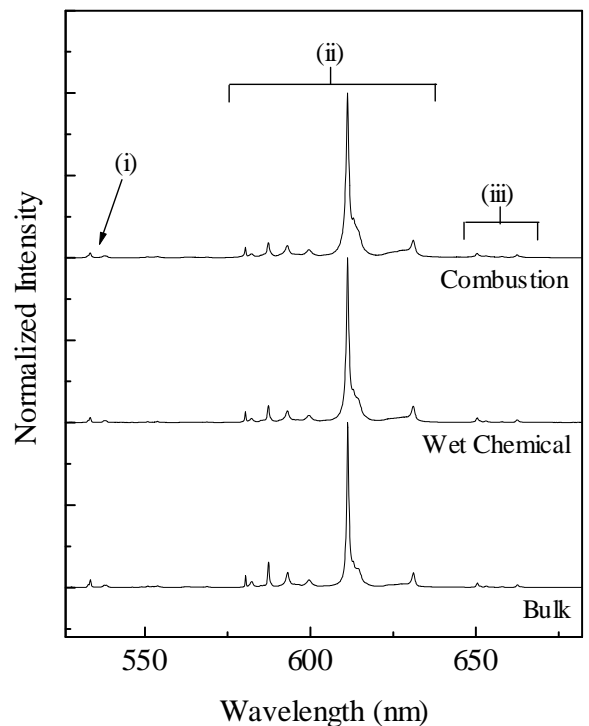


Figure 2: RT luminescence of Y₂O₃:Eu³⁺, Ho³⁺ samples upon excitation at 257.25 nm. (i) ⁵D₁ → ⁷F₀ [Eu³⁺] (ii) ⁵D₀ → ⁷F₀, ⁷F₁, ⁷F₂ [Eu³⁺] (iii) ⁵D₀ → ⁷F₃ [Eu³⁺]