

Capacitance probe of the electron displacement in a dye sensitised solar cell by an intermodulation technique : a quantitative model.

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For a dye-sensitised photo electrochemical cell (DSSC) SnO₂ / TiO₂ (nanocrystalline) / redox electrolyte /Pt/ SnO₂, it is well admitted that electrons injected from the excited state of the dye are diffusing through the TiO₂ nanostructure. As a consequence, the potential difference between the TiO₂ surface and the rear electrical contact essentially takes place at the SnO₂/TiO₂ junction. Impedance measurements show that doped SnO₂ is depleted at the SnO₂/TiO₂ interface in the potential range U= 0 (short circuit) to V_{oc} (open circuit). The corresponding space-charge differential capacitance C_d is found to present a Mott-Schottky behaviour, with a potential barrier maximum at U=0 and minimum at U=V_{oc}. For usual DSSC, C_d can be measured in the kHz range. Under illumination, electron concentration at the SnO₂/TiO₂ interface is defined by its Quasi-Fermi level (QFL), lined up with the Fermi level of SnO₂. When modulating the applied potential U at a frequency f₂ much lower than the C_d sampling frequency f₁, the capacitance C_d is modulated *via* the changes induced in the QFL by the time-dependent processes implied in the electron displacement in the TiO₂ structure. The so-called *intermodulation technique* consists of applying a composite sinusoidal signal ΔUf₁ + ΔUf₂ to the DSSC, with f₁ >> f₂. The quadrature component of the *ac* total current is detected at the frequency f₁ by a Lock-In amplifier. Under certain conditions, the output signal is proportional to the C_d capacitance with a residual modulation at f₂. The f₂ component is then determined by a Frequency Response Analyser, typically in the range f₁/10 > f₂ > f₁/10⁵ [1,2].

Dealing with capacitance, the charge-potential Q(U) characteristic is relevant to define the state of the electrochemical system. At any potential U, the ratio Q/U is defined as the integral capacitance C_{int}. The time-dependence of the displacement current I_d is I_d=dQ/dt. When superimposing a sinusoidal signal at ω₁=2πf₁ to the *dc* potential U_o, the component of the current I_d at f₁ is :

$$I_{f1} = [U_o(dC_{int}/dU) + C_{int}(U_o)].j\omega_1.\Delta U f_1$$

where the term in brackets is the differential capacitance C_d. In the case of a composite signal ΔUf₁ + ΔUf₂, it was demonstrated [2] that :

$$I_{f1f2} = [U_o(d^2C_{int}/d^2U) + 2(dC_{int}/dU_o)].j\omega_1.\Delta U f_1.\Delta U f_2$$

where I_{f1f2} represents the modulation at the frequency f₂ of the displacement current measured at f₁. The response of C_{int} to a potential perturbation is expressed as the sum of two terms, the first one is time-

independent, only potential dependent, the second one is time-dependent. The transfer function experienced by the inter modulation technique is written as :

$$\Delta C_{f1}(f_2)/\Delta U(f_2) = \{\Delta C_{f1}/\Delta U\}_U + \{\Delta C_{f1}(f_2)/\Delta U\}_t$$

with

$$\{\Delta C_{f1}/\Delta U\}_U = [U_o(d^2C_{int}/d^2U) + 2(dC_{int}/dU_o)]$$

For DSSC, a quantitative expression for C_{int} is obtained when considering the space-charge formed at the SnO₂/TiO₂ interface. It allows the frequency independent term to be calculated. It is expressed as :

$$\{\Delta C_{f1}/\Delta U\}_U = - (q\epsilon\epsilon_o N/8)^{0.5} . (U - U_{fb})^{-1.5}$$

where q is the elementary charge, εε_o and N are the dielectric permittivity and carrier density of SnO₂. U_{fb} is the flatband potential with respect to the SnO₂/TiO₂ interface. The frequency dependent term {ΔC_{f1}(f₂)/ΔU}_t is calculated using expressions developed for IMPS [3]. It is formally expressed as

$$\{\Delta C_{f1}(f_2)/\Delta U\}_t = A.th\sqrt{z} / (\sqrt{z} + B.th\sqrt{z})$$

where the complex argument z depends on the processes taken into account (diffusion, recombinations, trapping). Capacitance being proportional to the surface area, this new transfer function allows to determine the actual TiO₂ surface coverage of the SnO₂ substrate. If normalized with respect to the geometric surface area, the measured (real or complex) quantity is expressed in nF V⁻¹ cm⁻² units.

As a first example, in the dark, in the absence of any current, the response is a real constant. As a second example, an experimental Nyquist diagram recorded under illumination is given in the figure with comparison to the simulated curve according to the above model. The well separated low frequency loop is attributed to electron trapping-detrapping effects. The high frequency loop is related to diffusion-recombination processes.

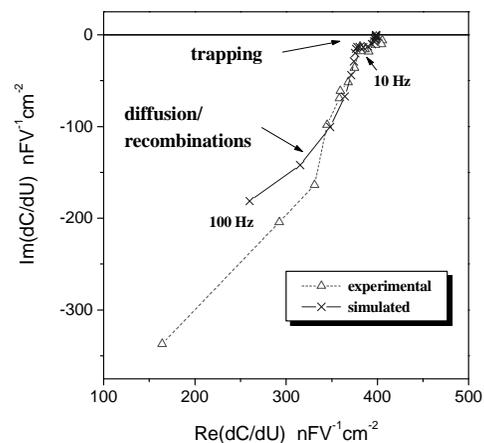


Fig.1: Comparison between experimental and simulated Nyquist diagrams for a DSSC cell using a solid PEO-based electrolyte (Diffusion coefficient = 3.10⁻⁴ cm².s⁻¹; N = 2.10²⁰ cm⁻³; trapping rate constant= 50 s⁻¹; surface area = 0.3 cm²).

References

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