

Cu, Fe, Co, AND Mn SILD TECHNOLOGY FOR SURFACE MODIFICATION OF SnO₂ GAS SENSITIVE FILMS

G.Korotcenkov¹, V.Macsanov¹, V.Brinzari¹,
V.Tolstoy², J.Schwank³, J.Morante⁴

¹Technical University of Moldova, Bld. Stefan cel Mare,
168, Chisinau, 2004, Rep. of Moldova,
E-mail: koro@ch.moldpac.md

²St. Petersburg State University, Russia;

³University of Michigan, Ann Arbor, MI, U.S;

⁴University of Barcelona, Barcelona, Spain;

In this report we discuss possibilities of SILD (successive ionic layer deposition) technology for surface modification of nano-scaled SnO₂ films by nano-clusters of Cu, Fe, Mn, and Co oxides. These elements are widely used for improvement of gas sensing characteristics of metal oxide gas sensors. Samples of SnO₂ films with thickness (30-50 nm) were deposited by spray pyrolysis [1].

Method of SILD consists essentially in repeated successive treatments of the substrate surface with solutions of various salts, anions and cations, which form poorly soluble compounds [2]. Then samples were washed with distilled water, which removed excess of salt; treated with the H₂O₂ (OH⁻) solution, and again washed. The duration of each treatment was varied from 0,25 to 2 min. One such treatment comprised one deposition cycle. The ellipsometry method has shown that after each cycle we can deposit the layer of peroxocomplexes of metals with thickness 0,6-1,5 nm. During our experiments we used from 1 to 8 deposition cycles. The aqueous of 0,01-0,001M solutions of Fe(NH₃)₂SO₄, CoSO₄, CuAc₂, and MnCl₂ (KMnO₄) were used as a precursors for synthesis of FeOOH, CoOOH, Cu(OH)_x(OOH)_{2-x}, and MnO₂xH₂O correspondingly. On relatively weak heating (100-400°C), hydroxides, peroxides and peroxide hydrates release water and transform into corresponding oxides. Gas response to CO, O₃ and H₂ was tested in steady state and transient modes in the range of operating temperatures equaled 50-500°C.

It was established that the influence of surface deposition of indicated oxides on both the electro-physical, and gas sensing characteristics of SnO₂ films takes place. I.e. this method of surface modification by SILD can be used for improvement of sensitivity, selectivity, and rate of SnO₂ gas response. This influence depends on the nature of metal oxide, size of cluster, and operating temperature. Surface modification by Cu, Co, Mn, and Fe increases SnO₂ film resistance. Addition of Cu depresses gas response to ozone. Modification by Co improves both the gas response to H₂ in the range of low operating temperatures, and the rate of gas response. Fe can be used for improvement of both gas response to CO, and selectivity. Surface modification by Mn depresses gas response to both CO, and H₂.

The comparison of gas sensing properties of SnO₂ films with bulk and surface modification by Co, Mn, Cu, and Co was conducted. The origin of surface modification influence on SnO₂ gas response is discussed.

References.

- [1]. G.Korotcenkov, V.Brinzari, M.DiBattista, J.Schwank, A.Vasiliev *Sensors and Actuators*. **B.77**(1-2) (2001) 244-252.
- [2]. V.P.Tolstoy, *Thin Solid Films*, **307** (1997) 10-13.

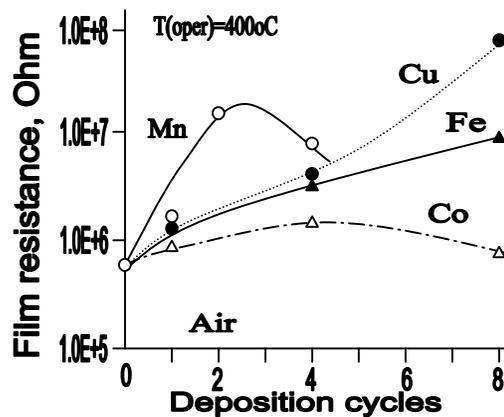


Fig.1. Surface modification influence on SnO₂ film resistance.

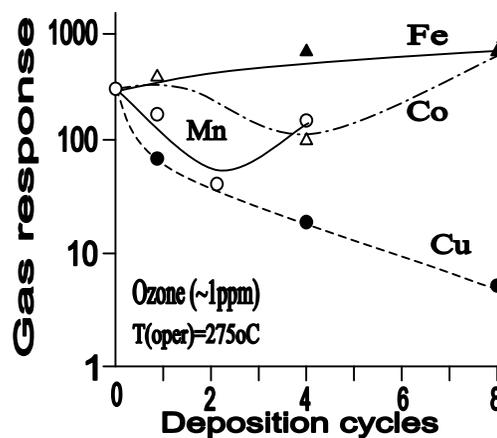


Fig.2. SnO₂ gas response to ozone vs. surface modification.

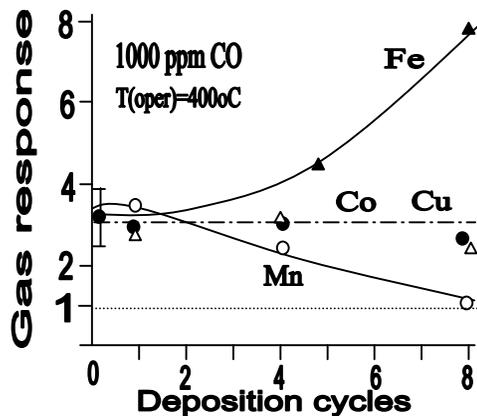


Fig.3. Influence of surface modification on gas response to CO

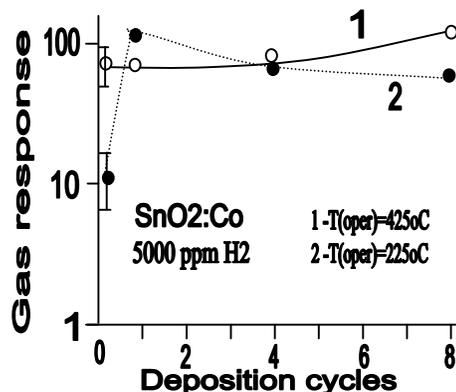


Fig.4. Gas response to H₂ in dependence on operating temperatures