PHOTOINDUCED PROCESSES OF -FULLERENE LINKED POLYCARBAZOLE IN SOLUTION

Osamu Ito, Tatsurou Midorikawa, Yasuyuki Araki, Mamoru Fujitsuka

Institute of Multidisciplinary Research for Advanced Materials, Tohoku University, Japan Science, Katahira, Sendai, Miyagi 980-8577, Japan.

Light induced photocurrent has been generated by the of fullerene (C_{60}) excitation containing (PVCz) films.[1] Photoinduced poly(vinylcarbazole) bimolecular electron transfer between C₆₀ and PVCz and oligocarbazole has been previously reported [2]. In the present study, photoinduced charge-separation (CS) and recombination (CR) processes of fullerene (C₆₀) pendant polycarbazoles (PVCz-C₆₀), which were designed to change the composition of C_{60} by copolymerization of C₆₀-styrene and VCz as shown in Scheme 1, have been investigated by time-resolved fluorescence and absorption spectroscopic methods The absorption spectra are shown in Fig. 1, in which the peak at 700 nm is attributed to the C₆₀ moiety; however, broad absorptions in the visble region indicate the charge-transfer interaction between the C_{60} and Cz moiety. By the selective excitation of the C_{60} moiety of PVCz-C₆₀ in benzonitrile (PhCN), chargeseparation takes place via the excited singlet state of C₆₀. The nanosecond transient spectra observed by the selective excitation of C₆₀ are shown in Fig. 2, in which the absorption of the radical cation of the carbazole moiety (Cz'+) appeared at 780 nm and the chargeresonance band of dimer radical cation of carbazole $(\text{Cz})_2`^+$ appeared at near-IR region, while the radical anion of C_{60} (C_{60}) appeared at 1000 nm, indicating that the charge-separated state $PVCz^+-C_{60}^-$ was produced via $PVCz^{-1}C_{60}^{*}$. The lifetimes of the charge-separated states (τ_{CS}) of $PVCz^{+}\text{-}C_{60}^{--}$ vary with the composition of C_{60} in PVCz-C₆₀; $\tau_{CS} = 2.80$ ms for PVCz-C₆₀ (1%), $\tau_{CS} = 0.43$ ms for PVCz-C₆₀ (3%), and $\tau_{CS} = 0.24$ ms for PVCz-C₆₀ (5%). These changes in the τ_{CS} values with the composition of C₆₀ in PVCz-C₆₀ afforded rich information about the flow of the electron and hole along the PVCz- $C_{60} \, as$ shown in Scheme 2. Simulation (Fig. 3) gave the rate constants as summerized in scheme 2. Variation of the compositions of $C_{60} \mbox{ in } PVCz\mbox{-}C_{60} \mbox{ affects the rate}$ parameters, which rationalized the observed τ_{CS} values and rate parameters of the flow of the electron and hole along the PVCz-C₆₀ appreciably.

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Figure 1. UV-Vis absorption spectra of PVCz- C_{60} (n%) (2mM, monomer unit) in THF.



Figure 2. Transient absorption spectra observed by 532 nm-laser irradiation of PVCz-C₆₀ (5%) (2 mM; monomer unit) in deaerated PhCN.



Figure 3. Time-profile of PVCz- C_{60} (1%) in benzonitrile at 1500 nm and calculated decay.