

Water-soluble hat-stacked-type carbon nanofibers for biomedical applications

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Nanocarbon materials have been widely studied as catalyst supports in heterogeneous catalysis because of their unique morphology and reactivity¹. Recently, a novel type of carbon nanofiber exhibiting cup-stacked morphology was developed as a unique and functional “tubular” nanosized materials². Generally speaking, the morphology of carbon nanofibers is dependant on the metal catalyst. We have synthesized the carbon nanofibers with the structure in which graphene-hats are stacked toward a needle-axis. Here, we call these “Hat-stacked-type carbon nanofibers (Hat-stacked-type CNFs)”. As the edges of the graphene-hat have been exposed, many hydrophilic groups can be added in order to dissolve them into water and to use them biomedical application such as a carrier of DNA transduction. Also, as each hat is van der Waals force in which binding force is weak, hat-stacked-type CNFs can be cut off, that is, it is possible to control their length size. Thus, hat-stacked-type CNFs can be expected as a biomaterial as well as carbon nanotubes³⁻⁶.

In this presentation, we report simple cutting and water-solubilization of the hat-stacked-type CNFs using sonication in a mixture of concentrated acids⁷. We purified the hat-stacked-type CNFs (Figure 1) and characterized the cut hat-stacked-type CNFs by SEM, TEM, XRD, and FT-IR measurements. The hat-stacked-type CNFs were easily cut into 400-nanometer to 1.5-micrometer lengths, and the cut nanofibers formed a stable solution in water.

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References

1. J. M. Planeix et al., *J. Am. Chem. Soc.* **116**, 7935 (1994).
2. M. Endo et al., *Appl. Phys. Lett.* **80**, 1267 (2002).
3. D. Pantarotto et al., *J. Am. Chem. Soc.* **125**, 6160 (2003).
4. F. Belavoine et al., *Angew. Chem. Int. Ed.* **38**, 1912 (1999).
5. A. Bianco and M. Prato, *Adv. Mater.* **15**, 1765 (2003).
6. D. Pantarotto et al., *Chem. Comm.* 16 (2004).
7. J. Liu et al., *Science* **280**, 1253 (1998).

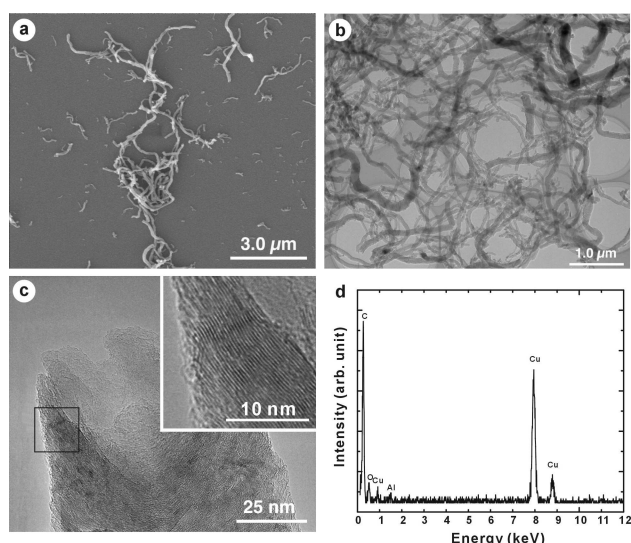


Figure 1. SEM photograph (a), TEM photographs (b and c) of the low and high magnification of the purified hat-stacked-type CNFs. Inset shows the lattice image of the graphene-hat layers. EDX spectrum (d) of the purified hat-stacked-type CNFs.