## Tuning of Work Function of Polyaniline Films for HCN Gas Sensors

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GasFET (Gas sensitive Field Effect Transistor) is a device that measures work function (WF) changes of the gate conductor when exposed to an analyte<sup>1</sup>. The WF changes are monitored by measuring the change in turnon voltage (threshold voltage) of the device. One of the unique aspects of the GasFET is that the response (or sensitivity) of the device is controlled by charge–transfer interactions between the analyte and the gate material<sup>1</sup>. Because of that, the tuning of the initial WF of the gate conductor is very important. In our approach we use polyaniline films that are cast from formic acid, electrochemically deposited and/or modified.

The PANI films were exchanged electrochemically for different carboxylic acids<sup>2</sup> (Fig.1) after the deposition in the presence of mercuric chloride that forms HCN sensitive binding sites<sup>3</sup>. To investigate the effect of the different carboxylic acids, we used FT-IR peaks<sup>2</sup> and turn-on voltage of the GasFETs. The cathodic peak around 0V (shown by '\*' in Fig. 1) was correlated with the quinoid to benzenoid peaks ratio from FT-IR (Fig. 2), and turn-on voltage (Fig. 3). Gas exposure experiments to HCN were conducted and the responses were correlated with the initial WF of the films obtained by turn-on voltage. Currently, optimization of the films is in progress, in order to improve the sensor performance.

Acknowledgement

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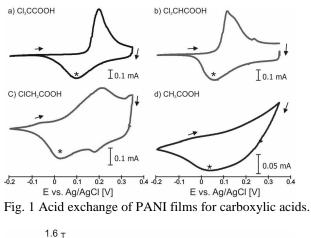
References

(1) Janata, J.; Josowicz, M. Accounts Chem Res **1998**, *31*, 241-248.

(2) Hatchett, D.; Josowicz, M.; Janata, J. Jornal of

physical chemistry B **1999**, 130, 10992-10998.

(3) Langmaier, J.; Janata, J. Anal Chem **1992**, 64, 523-527.



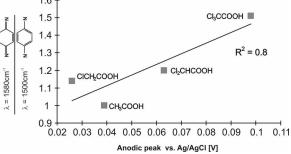


Fig.2 Correlation between quinoid to benzenoid ration and cathodic peak obtained din Fig. 1.

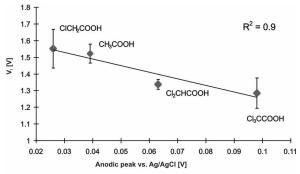


Fig.3 Correlation between turn-on voltage and cathodic peak obtained din Fig. 1.