

Sensitization of SnO₂-based thick film sensor for ethylene oxide gas

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Ethylene oxide (EO, C₂H₄O) is an important industrial chemical. Besides the same gas is an excellent disinfectant used popularly in hospital and food industry for disinfection of heat-sensitive materials and equipments. As recently recognized, the gas is hazardous to human beings [1,2], and its safety standard at working places has been regulated to be 1ppm in Japan and US. Thus a compact inexpensive EO gas sensor which can cover this standard is in great demand. This situation prompted us to try the possibility to develop such an EO gas sensor based on semiconducting oxides. Although SnO₂ turned out to be one of the most promising oxides for EO sensing, thick film devices using neat SnO₂ remained far from meeting the sensitivity required[3]. It was found further that loading SnO₂ with La₂O₃ was effective in promoting the sensitivity, and that the deterioration of response and recovery rate induced could be restored fairly effectively by additional loading with Pd or Pt. The doubly promoted device thus prepared, Pt(0.5wt.%)–La₂O₃(0.5wt.%)–SnO₂ showed sensor response (R_{air}/R_{gas}) exceeding 10 to 1 ppm EO as well as acceptable rates of response under dry condition at 300°C as shown in Figure 1. Because of the double promotion, however, electrical resistance of the device in air went up almost to reach the upper limit of reliable measurement (10⁸Ω) at this temperature(300 °C). This suggests it necessary to lower the electrical resistance for further explanation. In this study, we tried to dope SnO₂ with Sb for this purpose.

To prepare Sb-doped SnO₂, a solution mixed of SnCl₄ and SbCl₅ was added to an aqueous solution of NH₄HCO₃ for hydrolysis. The resulting precipitate, after through washing, was collected by filtration, dried at 120 °C, and calcined at 700 °C in air. The powder was loaded with additives by an impregnation method. To fabricate sensor devices, the powders were screenprinted on an alumina substrate attached with gold electrodes and calcined at 600 °C for 3h. Gas sensing properties were measured in a conventional gas flow apparatus in the temperature range of 250 – 500 °C.

Figure 2 shows the electrical resistances of the devices using neat and Sb-doped SnO₂ in air (R_a) and in 20 ppm EO containing air (R_g) as a function of operating temperature. The Sb doping resulted in drastic decreases in resistances, as seen from the comparison among the promoter-unloaded devices. In view of sensor devices, an optimum amount of Sb doping appears to be about 0.05at.%. As stated before, the resistances increased on loading these base oxides with La₂O₃(0.5wt.%). However, the resistance of the La₂O₃-loaded device using Sb(0.05at.%) doped SnO₂ is seen to be about 10⁶Ω yet at 250°C, although the resistance after further loading with Pt is to be measured.

Figure 3 shows sensor response to various concentrations of EO for the La₂O₃-loaded devices using neat and Sb(0.05at.%) doped SnO₂ at 250°C. The both devices could respond to 1-15ppm C₂H₄O fairly well. This confirms that the sensor response is not affected so much by the Sb doping. Further studies on Pt-La₂O₃-doubly loaded devices using Sb-doped SnO₂ are in progress.

References

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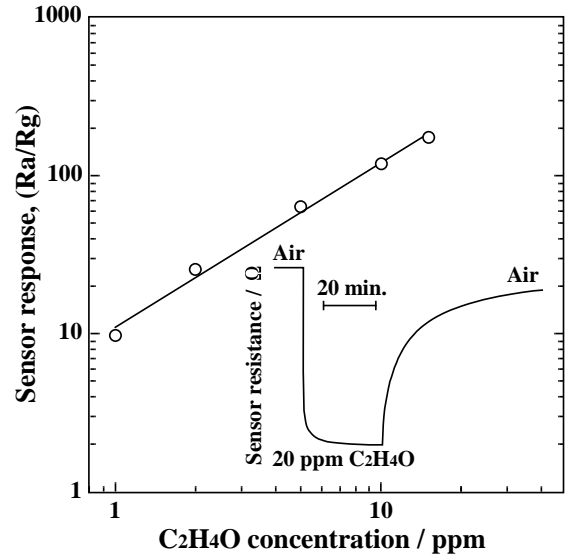


Fig.1 Sensor response to various concentrations of C₂H₄O and typical response transient for Pt(0.5wt.%)–La₂O₃(0.5wt.%)–doubly loaded-SnO₂ device at 300°C.

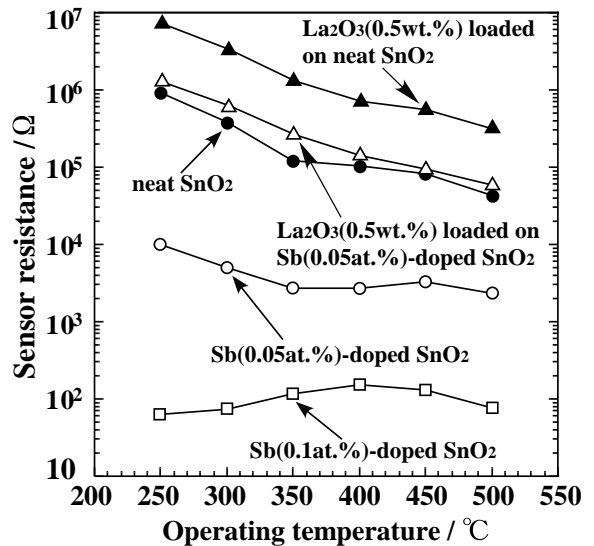


Fig.2 Electrical resistances vs. operating temperature for La₂O₃-loaded or unloaded devices using neat or Sb-doped SnO₂.

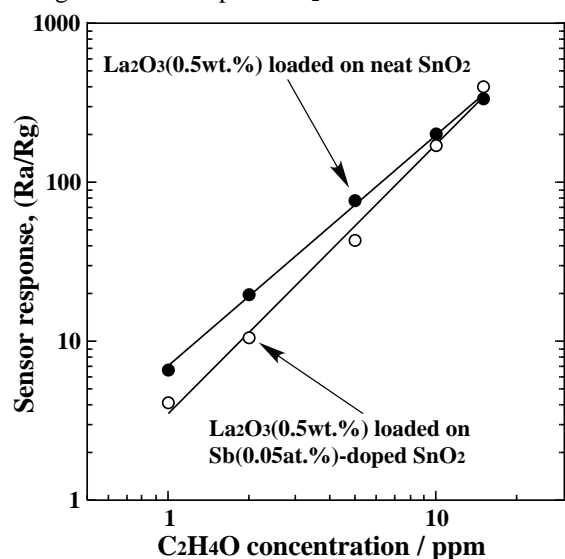


Fig.3 Sensor response to various concentrations of C₂H₄O for La₂O₃(0.5wt.%) loaded devices using neat and Sb-doped SnO₂ at 250°C.